## 142. Photochemical Reactions

148<sup>th</sup> Communication<sup>1</sup>)

## Photochemistry of Acylsilanes: Preparation, Photolyses, and Thermolyses of α,β-Unsaturated Silyl Ketones

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The syntheses, photolyses, and thermolyses of the  $\alpha$ , $\beta$ -unsaturated silyl ketones (E/Z)-7, (E)-8, and (E)-9 are described. On  $n,\pi^*$ -excitation ( $\lambda > 347$  mm), the aforementioned compounds undergo (E/Z)-isomerization followed by  $\gamma$ -H abstraction. The intermediate enols are trapped intermolecularly by siloxycarbenes leading to the dimeric acetals 27A + B, 30A + B, and 31A + B. In addition, the acylsilanes (E/Z)-7 undergo photoisomerization by  $\delta$ -H abstraction furnishing the acylsilanes 29A + B. Flash vacuum thermolyses (FVT) of (E/Z)-7, (E/Z)-8, and (E)-9 give rise to intramolecular reactions of the siloxycarbene intermediates. Thus, FVT (520°) of (E)- and (Z)-7 selectively leads to the enol silyl ethers 32 and (E)-33, respectively, arising from carbene insertion into an allylic C-H bond. FVT of (E/Z)-8 (560°) and (E)-9 (600°) affords the trienol silyl ethers 34A + B and the cyclic silyl ethers 37A + B, respectively, which are formed by CH insertion of the siloxycarbene. As further products of (E)-8 and (E)-9, the bicyclic enol ethers 35 and 36 are formed, presumably *via* siloxycarbene addition to the cyclohexene C=C bond.

1. Introduction. – Recently, we reported on the photochemistry of the acylsilane 1 [3]. It was found that the major photochemical processes are *Norrish* type-II fragmentation (leading to the diene 2 and the enol 3) and rearrangement to the siloxycarbene **a** (*Scheme 1*). The latter reacts by *intermolecular* insertion of the carbene center into the O-H bond of 3 (furnishing the enol acetal 4) and by *intramolecular* addition of the carbene center to the methylidene group or insertion into an adjacent C-H bond (leading to compounds 5 and 6, respectively, as minor products).

As part of our studies of the intramolecular trapping of siloxycarbenes by reaction with various neighboring groups, we describe here the photolyses and thermolyses of the  $\alpha,\beta$ -unsaturated acylsilanes (E/Z)-7, (E)-8, and (E)-9 (Scheme 2). They were considered as suitable models to delineate the reactivity of the expected siloxycarbene intermediate with a neighboring double bond. It was of particular interest to find out, whether or not the siloxycarbene intermediate would undergo ring closure to a cyclopropene, as observed in vinyl carbenes [4].

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2. Preparation of the Acylsilanes (E/Z)-7, (E)-8, and (E)-9. – The compounds (E/Z)-7, (E)-8, and (E)-9 were synthesized via the dithiane route<sup>5</sup>) starting from the aldehydes (E/Z)-10, 11, and 12, respectively. Thus, thioacetalization of citral (E/Z)-10 (ca. 2:1) with propanedithiol  $(BF_3 \cdot Et_2O, CH_2Cl_2)$  furnished the 1,3-dithianes (E/Z)-13 [6] (ca. 3:2; 91%), which were transformed to (E/Z)-14 (ca. 3:2; 98%) by reaction with BuLi and (t-Bu)Me<sub>2</sub>SiCl. Dethioacetalization with Tl(NO<sub>3</sub>)<sub>3</sub>·3H<sub>2</sub>O [7] led to the acylsilanes (E/Z)-7 (ca. 2:1; 50%); overall yield 45%)<sup>6</sup>).



<sup>&</sup>lt;sup>5</sup>) After completion of this work [2], a further practical and efficient synthesis of  $\alpha,\beta$ -unsaturated acylsilanes involving silyl-*Wittig* rearrangement and *Swern* oxidation was published by *Danheiser et al.* [5].

<sup>&</sup>lt;sup>6</sup>) Dethioacetalization of pure (E)- or (Z)-14 under the same conditions also led to a ca. 2:1 mixture of (E)/(Z)-7.

Analogously, the acylsilanes (E)-8 and (E)-9 were obtained from the aldehydes 11 (via 15 and 16) and 12 [8] (via 17 and 18), respectively. Compounds 11 and 12 were prepared by degradation of 8-methyl- $\alpha$ -ionone (19) and  $\alpha$ -ionone (20), respectively, (a) I<sub>2</sub>/pyridine, b) LiAlH<sub>4</sub>) and subsequent oxidation of the alcohols 21 and 22 [9] with MnO<sub>2</sub>.

It is noteworthy that dethioacetalization of 16 and 18 did not cause  $(E \rightarrow Z)$  isomerization as was observed for (E)-14. Furthermore, using HgO/HgCl<sub>2</sub> [10] instead of Tl(NO<sub>3</sub>)<sub>3</sub>·3H<sub>2</sub>O for the cleavage of the dithiane 16, the yield of the acylsilane (E)-8 improved from 45% to 93%<sup>7</sup>).

**3.** Photolyses of (E/Z)-7, (E)-8, and (E)-9. – 3.1. Irradiation of (E)-7 (0.010M soln. in MeCN;  $\lambda > 347$  nm; 100% conversion) afforded: **27A** + **B**<sup>8</sup>) (ca. 1:1; 22%), **28** (12%), **29A** + **B** (ca. 1:1; 10%; Scheme 4), and (E/Z)-10 (17%; Scheme 2).

3.2. Irradiation of (Z)-7 (0.015M soln. in MeCN;  $\lambda > 347$  nm; 83% conversion) gave **27A** + **B** (*ca.* 1:1; 24%), **28** (19%), **29A** + **B** (*ca.* 1:1; 14%), and (*E*/Z)-**10** (16%).



<sup>7</sup>) An analogous improvement was observed on dethioacetalization of the dithianes 23 and 24 [2] (Scheme 3). With Tl(NO<sub>3</sub>)<sub>3</sub> · 3 H<sub>2</sub>O, the acylsilanes 25 and 26 were produced in only ca. 25% yield, whereas the reaction with HgO/HgCl<sub>2</sub> led to the yields of 86 and 88%, respectively. On dethioacetalization of (E/Z)-14→(E/Z)-7 with HgO/HgCl<sub>2</sub> instead of Tl(NO<sub>3</sub>)<sub>3</sub> · 3 H<sub>2</sub>O, however, the yield could be increased only slightly from 50 to 56%.



<sup>8</sup>) The terms **A** and **B** are used for the description of diastereoisomers whose configurations were not assigned conclusively.

3.3. Irradiation of (E)-8 (0.012M soln. in MeCN;  $\lambda > 347$  nm; 98% conversion) gave (Z)-8 (21%), 30A (4%), and 30B (3%; Scheme 4). Irradiation of (E)-8 in THF under the same conditions afforded (Z)-8 (22%), 30A (6%), and 30B (4%).

3.4. Irradiation of (E)-9 ( $\lambda > 347$  nm) in MeCN or THF led to mixtures containing *ca*. 30% of the acetals **31A** + **B** which, however, decomposed on chromatography on SiO<sub>2</sub>.

4. Thermolyses of (E/Z)-7, (E/Z)-8, and (E)-9. – 4.1. Flash Vacuum Thermolysis (FVT [11]) of (E)-7 and (Z)-7 (520°) afforded the dienol ethers 32 (42%) and (E)-33 (46%), respectively (Scheme 5).



4.2. FVT of (E)- and (Z)-8 (560°) furnished the trienol ethers 34A (17 and 15%, respectively) and 34B (5%), and the enol ether 35 (30%; Scheme 5).

4.3. *FVT* of (*E*)-9 (600°9)) gave compounds 36 (25%), and 37A + B (*ca.* 1:1; 39%; *Scheme 5*).

5. Structure of the Products. – The structures of all new compounds were deduced from the spectral data, of which only the most relevant are discussed herein together with the chemical transformations which confirmed the assigned structures. Full data and the assignment of the NMR data are presented in the *Exper. Part.* 

 $\alpha,\beta$ -Unsaturated Silyl Ketones (E/Z)-7, (E/Z)-8 and (E)-9. In particular, these compounds show IR bands shifted to long wavelengths: 1625–1630 cm<sup>-1</sup> and 1560–1595 cm<sup>-1</sup>. In the UV spectra, apart from the strong  $\pi,\pi^*$  bands between 225–260 nm ( $\epsilon \approx 10000$ ), the characteristic structured  $n,\pi^*$  bands in the region of 400–490 nm ( $\epsilon \approx 100$ ) are observed. Furthermore, in the <sup>13</sup>C-NMR spectra, the C=O groups are evidenced by the low-field shifted signals at *ca*. 235 ppm.

<sup>&</sup>lt;sup>9</sup>) This temperature was necessary to convert 75-80% of the starting material.

Enol Acetals 27A + B, 30A + B, and 31A + B (Scheme 4). The UV maximum of 27A + B at 233 nm ( $\varepsilon = 7600$ ) indicates the dienolether moiety. Characteristic <sup>1</sup>H-NMR signals are at 2.70 ppm (*tm*) of the double allylic 2H-C(4), at *ca*. 5.35 ppm (*dm*, J = 7 Hz) of H-C(11), and at *ca*. 5.80 ppm (*d*, J = 7 Hz) of the acetal H-C(10). In the <sup>13</sup>C-NMR spectrum, a *d* at 93 and a *s* at 158 ppm are characteristic of the acetal C(10) and the enolether C(8), respectively. Conclusive evidence for the structure of 27A + B was obtained on hydrolysis (2m HCl/Et<sub>2</sub>O) leading to the acylsilanes (*E*/*Z*)-7 and 28, and to the aldehydes (*E*/*Z*)-10 (Scheme 2). The enol acetals 30A and 30B, which were separated by column chromatography, show spectroscopic data similar to those of 27A + B (see Exper. Part). Hydrolysis of 30A and 30B (aq. HCl/Et<sub>2</sub>O) afforded the acylsilane (*Z*)-8 and the aldehyde (*E*)-11 (Scheme 2).

Hydrolysis of 31A + B (THF/5% aq. HCl) gave the aldehyde 12 as the only product. On treatment of the acylsilanes (*E*)-8 and (*E*)-9 under the same conditions, it was shown that (*E*)-8 was stable, whereas (*E*)-9 was transformed to the aldehyde 12.

Acylsilanes 28 and 29A + B (Scheme 4). These compounds show spectral data characteristic of acylsilanes (see *Exper. Part*). Furthermore, the structure was assigned by comparison of their spectral data with those of the related methyl and phenyl ketones [12].

The Dienol Silyl Ethers **32** and (E)-**33** show strong IR bands (1635 and 1640 cm<sup>-1</sup>, respectively) and UV maxima at *ca*. 240 nm ( $\varepsilon \approx 20000$ ), which are characteristic of the dienol-ether moieties. In the <sup>1</sup>H-NMR spectra, the coupling constants J = 12.5 Hz of the *AB* systems at *ca*. 6.4 ppm indicate the (E)-configuration around the C(1)=C(2) bond. The (E)-configuration around the C(3)=C(4) bond of (E)-**33** was assigned by an NOE experiment: on irradiation at 2.84 ppm (2H-C(5)), the signals of CH<sub>3</sub>-C(3) and CH<sub>3</sub>-C(7) showed a positive, and the *AB* system of H-C(1) and H-C(2) a negative NOE [13]. As structure proof, the dienol ethers **32** and (E)-**33** were obtained together with their isomer (Z)-**33** (see below) by reaction of (E/Z)-**10** with (*t*-Bu)Me<sub>2</sub>SiCl and Et<sub>3</sub>N (92% combined yield). The (Z)-configuration around the C(3)=C(4) bond of (Z)-**33** was again assigned by an NOE experiment. Thus, irradiation at 2.88 ppm (2H-C(5)) caused a positive NOE of the signals of CH<sub>3</sub>-C(7) and H-C(2).

Trienol Silyl Ethers **34A** and **34B** (Scheme 5). The trienol-silyl-ether moiety is evidenced by the UV maxima at 221 nm ( $\varepsilon = 9390$ ) and 267 nm ( $\varepsilon = 9555$ ) of **34A**, by the IR bands at 1645 cm<sup>-1</sup> as well as by the <sup>1</sup>H- and <sup>13</sup>C-NMR-signals of the olefinic H- and C-atoms, respectively (see *Exper. Part*).

*Bicyclic Enol Ethers* **35** and **36** (*Scheme 5*). The assigned structures were confirmed by hydrolysis (5% aq. HCl/THF) leading to the bicyclic compounds **38** and **39** (see below) with a cyclopentanone moiety (IR: 1730 cm<sup>-1</sup>). For the NMR data, see *Exper. Part.* 

Cyclic Silyl Ethers 37A + B (Scheme 5). On treatment of 37A + B with 5% aq. HCl in THF, the silanol 40 incorporating a conjugated diene system was formed by hydrolysis and subsequent H<sub>2</sub>O elimination.



6. Discussion. -6.1. Photolyses.  $n,\pi^*$  Excitation ( $\lambda > 347$  nm) of the acylsilane (E)- or (Z)-7 causes rapid (E/Z)-isomerization. Thus, after a short time of irradiation, a photostationary equilibrium of (E/Z)-7 (ratio ca. 2:1) was obtained. As further main process,  $\gamma$ -H abstraction in (Z)-7 leads – via the diradical **b** – to the enol **41** (Scheme 6), which tautomerizes either to (Z)-7 or to the deconjugated acylsilane **28** (Scheme 4). The alternative deconjugated ketone **42**, which would be formed by photoenolization from (E)-7 ( $\rightarrow$  **c**  $\rightarrow$  **43**; Scheme 6), was not detected. In contrast to the dienol **41**, **43** can presumably more readily adopt the conformation necessary for a [1,5] signatropic H shift. Thus, reketonization to (E)-7 occurs instead of deconjugation to **42**<sup>10</sup>).

<sup>&</sup>lt;sup>10</sup>) For a recent investigation of the mechanisms of the photochemical enolization of  $\alpha$ . $\beta$ -unsaturated ketones, see [14].



 $R = CH = C(CH_3)_2$ 

The siloxycarbene **d** is formed from (E/Z)-7 via a second photoprocess, typical of acylsilanes [15]. Analogously to  $\mathbf{a} + \mathbf{3} \rightarrow \mathbf{4}$  (Scheme 1), the siloxycarbene **d** reacts with the dienol **41** leading to  $\mathbf{27A} + \mathbf{B}^{11}$ ). It was observed, however, that the formation and trapping of the siloxycarbene **d** is not as efficient as that of **a**. On photolysis of (E/Z)-7 in the presence of (t-Bu)Me<sub>2</sub>SiOH (1.5 equiv.), the acetals (E/Z)-45 (24%) are obtained together with **28** (13%) and **29A** + **B** (8%). On the other hand, on photolysis of **1** in the presence of (t-Bu)Me<sub>2</sub>SiOH (1 equiv.), the acetal corresponding to (E/Z)-45 is formed as the only product [3].

The cyclopropyl-acylsilanes 29A + B are most likely formed by  $\delta$ -H abstraction  $((Z)-7 \rightarrow e)$  followed by cyclization and ketonization  $(e \rightarrow 44 \rightarrow 29A + B)$ ; Scheme 6).

Comparison of the photoproducts of the  $n,\pi^*$  excitation of the silyl ketones (E/Z)-7 with those of the corresponding methyl ketones (E/Z)-46 [12] (Scheme 7) shows only the cyclopropyl compounds 29A + B and 47A + B as common product types. Thus, it is noteworthy that the methyl ketones (E/Z)-46 do not undergo photodeconjugation to 48, whereas related methyl ketones [16] with different substitution patterns as well as the phenyl ketone corresponding to (E/Z)-46 do show such a process [12]. On the other hand, the methyl ketones (E/Z)-46 undergo [2 + 2] cycloaddition and photo-ene reaction furnishing compounds 49 and 50 in high yield (Scheme 7), whereas an analogous photoprocess is not observed with the silyl ketones (E/Z)-7. This different behaviour may be explained by different reactivities of the excited states of silyl and methyl ketones.

<sup>&</sup>lt;sup>11</sup>) Products analogous to 27A + B, arising from reaction of d with 43 or 44, were not detected.



Previously, it was shown that the cyclization products 49 and 50 are formed *via* the  ${}^{3}n,\pi^{*}$  excited state of (E/Z)-46 [12]. On the other hand, it was disclosed that the  ${}^{3}n,\pi^{*}$  state of acylsilanes leads to siloxycarbene formation [17].

An analogous difference is also observed between the photolysis of the acylsilane (E)-8 and that of the corresponding methyl ketone 19 (*Scheme 2*) [18]. Analogously to (E/Z)-7, (E)-8 undergoes rapid (E/Z)-isomerization as well as  $\gamma$ -H abstraction and siloxycarbene formation leading to the enol acetals 30A + B (*Scheme 4*). n, $\pi$ \* Excitation of 19, however, gives rise to the formation of several cyclized products [18].

6.2. Thermolysis (520°) of (E)- and (Z)-7 selectively leads to the enol ethers 32 and (E)-33, respectively (Scheme 5). This transformation presumably involves siloxycarbene formation, without preceding (E/Z)-isomerization, followed by insertion of the carbene center into the allylic C-H bond ((E)-d  $\rightarrow$  51, (Z)-d  $\rightarrow$  52; Scheme 8)<sup>12</sup>). Although the cyclobutenes 51 and 52 are plausible intermediates, it may not be ruled out that 32 and (E)-33 are directly formed from the siloxycarbenes (E)- and (Z)-d. The fact that compounds 32 and (E)-33 are not formed by photolysis of (E/Z)-7 confirms our previous observation on comparison of the thermal and photochemical behaviour of the acylsilane



<sup>12</sup>) A similar reaction was previously found by *Shih* and *Swenton* [19]. Thermolysis of the acylsilane **53** afforded the aldehyde **54**. The authors assumed that the siloxycarbene **f** undergoes insertion into the C-H bond leading to the benzocyclobutene **55**, which undergoes further thermal rearrangements *via* **56**.



1 [13] (Scheme 1). Consequently, the photochemically generated siloxycarbenes **a** as well as **d** react preferentially by intermolecular insertion into an O–H bond rather than by an intramolecular insertion into a C–H bond or by addition to a C=C bond<sup>13</sup>). Furthermore, it is noteworthy that, on thermolysis of (E)- and (Z)-7, the  $\beta$ , $\gamma$ -unsaturated acylsilane **28** could not be detected, whereas the thermolysis of citral (E/Z)-10, the aldehyde related to (E/Z)-7, led to two  $\beta$ , $\gamma$ -unsaturated aldehydes [20].

Thermolysis of the acylsilane (*E*)-8 gives rise to (*E*/*Z*)-isomerization followed by carbene insertion into the C( $\gamma$ )-H bond furnishing the enol ethers 34A + B<sup>14</sup>). As additional product, the bicyclic enol ether 35 was obtained. Its formation can be explained by an intramolecular addition of the carbene center to the cyclohexene double bond (g  $\rightarrow$  57)<sup>15</sup>). The tricyclic compound 57, incorporating a vinylcyclopropane moiety, is assumed to be thermally unstable undergoing a [1,5]-homosigmatropic H shift [21] to 35.

On thermolysis of (E)-9, in addition to the bicyclic enol ether 36 (formed via  $h \rightarrow 58$ ), the cyclic silvl ethers 37A + B were obtained. The latters are obviously formed by siloxycarbene insertion into the C-H bond of the *t*-Bu group (Scheme 10)<sup>16</sup>).



7. Conclusion. – On  $n,\pi^*$  excitation, the  $\alpha,\beta$ -unsaturated silyl ketones (E/Z)-7, (E)-8, and (E)-9 show as main processes (E/Z)-isomerization,  $\gamma$ -H-abstraction, and formation of siloxycarbenes. The siloxycarbenes rapidly react with the enol intermediates (arising from  $\gamma$ -H abstraction) leading to dimeric acetals (see 27A + B, 30A + B, and 31A + B; Scheme 4). Products of intramolecular reactions of the photochemically generated siloxy-carbenes were, however, not detected. On the other hand, the thermally formed siloxy-carbenes react by insertion into C-H bonds or by addition to a C=C bond, thus showing reactivity typical of carbenes. In particular, these types of processes are also observed in vinyl carbenes [4]. As further reactions, vinyl carbenes undergo ring closure to cyclopropenes as well as insertion into adjacent olefinic C-H bonds leading to allenes [4]. Compounds of this type were, however, not detected as products of the unsaturated siloxycarbenes **d**, **g**, and **h**. On the basis of the aforementioned findings, it may be concluded that the unsaturated siloxycarbenes differ from vinyl carbenes. Furthermore, it may be suggested that the photochemically and the thermally generated siloxycarbenes are two differently reactive species.

<sup>&</sup>lt;sup>13</sup>) Shih and Swenton [19] found that the acylsilane 53 was photochemically quite stable. Therefore, they assumed that, if the carbene f is generated photochemically, it apparently has insufficient thermal energy to insert into the benzylic C-H bond.

<sup>&</sup>lt;sup>14</sup>) In the thermolysis mixture of (E)-8 (80% conversion), (Z)-8 could not be detected, however, thermolysis of (Z)-8 gave the same product distribution.

<sup>&</sup>lt;sup>15</sup>) An intramolecular addition of a thermally generated siloxycarbene to an electron-rich double bond was previously observed on FVT of the acylsilane 1 ( $\mathbf{a} \rightarrow 5$ ; Scheme 1).

<sup>&</sup>lt;sup>16</sup>) An analogous reaction was observed previously on thermolysis of pivaloyltrimethylsilane [22].

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## **Experimental Part**

General. See [3]. <sup>1</sup>H-NMR spectra were taken in CDCl<sub>3</sub> solns. or, exceptionally (as indicated below), in C<sub>6</sub>D<sub>6</sub> solns. on a *Bruker WP-80 CW* (80 MHz) or on a *Bruker WM 300* (300 MHz) instrument. *Filter A* 12. *Lamp B*: Hg medium-pressure lamp (125 W). Thermolyses were carried out in a quartz tube (30 cm  $\times$  1.7 cm), which was generally filled with quartz rings, and which was previously silylated by evaporation of bis(trimethylsilyl)-acetamide, at 0.5 Torr under N<sub>2</sub>. For a detailed description of the thermolysis apparature and procedure, see [11].

**1.** Preparations of (E/Z)-7, (E)-8, and (E)-9. – 1.1. Acylsilanes (E/Z)-7. 1.1.1. Thioacetalization of (E/Z)-10. To a soln. of (E/Z)-10 (11.76 g, 77.25 mmol) in abs. CH<sub>2</sub>Cl<sub>2</sub> (138 ml) and AcOH (*Fluka*, 138 ml) was added a soln. of propanedithiol (9.2 ml, 91.5 mmol) in AcOH (46 ml) and BF<sub>3</sub>·Et<sub>2</sub>O (6.9 ml, 54.9 mmol) at 0° during 15 min. After stirring for 3 h at r.t., the mixture was worked up in CH<sub>2</sub>Cl<sub>2</sub> (1000 ml) by the addition of ice and extraction with 2N aq. NaOH, 0.5N aq. NaHCO<sub>3</sub> and sat. aq. NaCl soln. Distillation (165°/0.1 Torr) afforded (E/Z)-13 [6] (17.1 g, 91%; 3:2 mixture).

(E)-2-(2',6'-Dimethyl-1',5'-heptadienyl)-1,3-dithiane ((E)-13). B.p. 165°/0.1 Torr. UV (1.731 mg in 20 ml): 235 (1350), 244 (sh), (1260), end absorption to 330. IR : 2970s, 2930s, 2890s, 2850s, 2825m (sh), 1655m, 1440m, 1430m, 1420m, 1410m, 1380m, 1375m, 1270s, 1240w, 1170m, 1165m, 1110w, 1050w, 1040w, 910m, 860m. <sup>1</sup>H-NMR (300 MHz): 1.59, 1,68, 1.75 (3d, J = 0.3, 1, 1.3, CH<sub>3</sub>-C(2'), CH<sub>3</sub>-C(6'), 3 H-C(7')); 1.77-2.20 (m, 2 H-C(5)), 2 H-C(3'), 2 H-C(4')); 2.75-2.86, 2.88-3.00 (2m, split into dddd, J = 14.5, 12, 3, 2.5, 2 H-C(4), 2 H-C(6)); 4.87 (d, J = 10, H-C(2)); 5.06 (m, J = 6,  $w_{y_{1}} = 5$ , H-C(5')); 5.14 (dm, J = 9.5,  $w_{y_{2}} = 4$ , H-C(1')). <sup>13</sup>C-NMR: 16.8, 17.7, 25.7 (3q, CH<sub>3</sub>-C(2'), CH<sub>3</sub>-C(6')), 25.1, 26.3, 39.3 (3t, C(5), C(3')), C(4')); 30.6 (t, C(4), C(6)); 44.4 (d, C(2)); 121.2, 123.6 (2d, C(1'), C(5')); 131.7, 140.8 (2s, C(2'), C(6')). MS: 242 (20,  $M^{++}$ , C<sub>13</sub>H<sub>22</sub>S<sub>2</sub>), 195 (13), 173 (54), 136 (14), 135 (55), 134 (12), 125 (19), 121 (16), 119 (21), 111 (11), 106 (13), 99 (46), 98 (13), 97 (14), 93 (34), 91 (11), 85 (10), 79 (11), 73 (10), 69 (97), 67 (25), 65 (13), 55 (13), 53 (14), 45 (21), 43 (16), 41 (100). Anal. calc. for C<sub>13</sub>H<sub>22</sub>S<sub>2</sub> (242.44): C 64.40, H 9.15; found: C 64.45, H 9.16.

(Z)-2-(2', 6'-Dimethyl-1', 5'-heptadienyl)-1,3-dithiane ((Z)-13). B.p. 165°/0.1 Torr. UV (0.136 mg in 2 ml): 235 (1120), 246 (960). IR: 2950s, 2920s, 2870s, 2850s, 1655w, 1465s (sh), 1455s, 1420m, 1410m, 1375s, 1275m, 1240w, 1170w, 1030w, 910m, 860m. <sup>1</sup>H-NMR (300 MHz): 1.63, 1.70, (2m,  $w_{1/2} = 3$ ) and 1.75 (d, J = 1.5, CH<sub>3</sub>-C(2'), CH<sub>3</sub>-C(6'), 3 H-C(7')); 1.76-1.89, 2.04-2.14 (2m, 2 H-C(5), 2 H-C(3'), 2 H-C(4')); 2.75-2.86, 2.88-3.00 (2m, split into dddd, J = 15, 12, 3, 2.5, 2 H-C(4), 2 H-C(6)); 4.87 (d, J = 10, H-C(2)); 5.16 (dm, overlapping with m, J = 10, H-C(1'), H-C(5')). <sup>13</sup>C-NMR (75 MHz): 17.7, 23.4, 25.7 (3q, CH<sub>3</sub>-C(2'), CH<sub>3</sub>-C(6'), C(7')); 25.0, 26.8, 32.7 (3t, C(5), C(3'), C(4')); 30.6 (t, C(4), C(6)); 44.2 (d, C(2)); 121.7, 123.8 (2d, C(1'), C(5')); 132.0, 141.2 (2s, C(2'), C(6')). MS: 242 (24,  $M^{++}$ , C<sub>13</sub>H<sub>22</sub>S<sub>2</sub>), 195 (17), 173 (52), 136 (16), 135 (67), 134 (15), 125 (18), 121 (20), 119 (43), 111 (13), 107 (12), 106 (14), 99 (49), 98 (14), 97 (15), 93 (44), 91 (13), 85 (10), 79 (13), 77 (11), 73 (11), 69 (90), 67 (25), 65 (15), 55 (14), 53 (15), 47 (10), 45 (23), 43 (20), 41 (100). Anal. calc. for C<sub>13</sub>H<sub>22</sub>S<sub>2</sub> (242.44): C 64.40, H 9.15; found: C 64.43, H 9.24.

1.1.2. Transformation of (E/Z)-13 to (E/Z)-14. To a soln. of (E/Z)-13 (9.31 g, 38.4 mmol, 3:2 mixture) in abs. THF (220 ml) and abs. HMPT (8 ml) was added a soln. of BuLi (28.9 ml, 1.6m in hexane, 46.2 mmol) at  $-78^{\circ}$ . After stirring the mixture for 1 h at  $-30^{\circ}$ , it was again cooled to  $-78^{\circ}$ , and a soln. of (t-Bu)Me<sub>2</sub>SiCl (6.97 g, 46.2 mmol) in abs. THF (42 ml) was added. The mixture was allowed to warm to r.t., stirred for 2 h, and worked up affording (E/Z)-14 (13.44 g, 98%; 3:2 mixture).

(E)-2-[(tert-Butyl)dimethylsilyl]-2-(2',6'-dimethyl-1,5-heptadienyl)-1,3-dithiane ((E)-14). B.p. 150°/0.07 Torr. UV (2.300 mg in 20 ml): 257 (730). IR: 2960s, 2930s, 2900s, 2850s, 1465m, 1460m, 1455m, 1440m (br.), 1430m, 1420m, 1405m, 1390m, 1380m, 1370m, 1360m, 1270m, 1255s, 1245s, 1190w, 1170w, 1150w, 1105w, 1070w, 1005w, 930w, 920m, 890w, 880w, 870w. <sup>1</sup>H-NMR (300 MHz): 0.16 (s, 2 CH<sub>3</sub>Si); 1.03 (s, 3 CH<sub>3</sub>CSi); 1.62, 1.68 (2m,  $w_{1/2} = 4$ , CH<sub>3</sub>-C(6'), 3 H-C(7')); 1.99 (d, J = 1.2, CH<sub>3</sub>-C(2')); 1.94-2.03, 2.06 · 2.18, 2.41-2.48, 3.02-3.12 (4m, 2 H-C(4), 2 H-C(5), 2 H-C(6), 2 H-C(3'), 2 H-C(4')); 5.13 (tm, J = 7,  $w_{1/2} = 4$ , H-C(5')); 5.57 (m,  $w_{1/2} = 5$ , 2.41-2.48, 2 H–C(1')). <sup>13</sup>C-NMR (75 MHz):  $-6.8 (q, 2 \text{ CH}_3\text{Si})$ , 15.2, 17.9, 26.0<sup>17</sup>) (3q, CH<sub>3</sub>–C(2'), CH<sub>3</sub>–C(6'), C(7')); 28.4 (q, 3 CH<sub>3</sub>CSi); 25.8, 26.0<sup>17</sup>), 26.8, 42.7, (5t, C(4), C(5), C(6), C(3'), C(4')); 124.6, 128.0 (2d, C(1'), C(5')); 19.8 (s, CSi); 45.7 (s, C(2)); 131.5, 137.0 (s, C(2'), C(6')). MS: 356 (3,  $M^+$ , C<sub>19</sub>H<sub>36</sub>S<sub>2</sub>Si), 223 (21), 193 (12), 191 (29), 149 (28), 115 (14), 106 (12), 91 (18), 74 (13), 73 (100), 69 (42), 59 (34), 44 (20), 41 (42). Anal. calc. for C<sub>19</sub>H<sub>36</sub>S<sub>2</sub>Si (356.71): C 63.98, H 10.17; found: C 64.17, H 10.06.

(*Z*)-2-[(tert-*Butyl*)dimethylsilyl]-2-(2',6'-dimethyl-1',5'-heptadienyl)-1,3-dithiane ((*Z*)-14). B.p. 150°/0.07 Torr. UV (2.247 mg in 20 ml): 257 (sh), (710). IR: 2960s, 2920s, 2900s, 2850s, 1725w, 1465m, 1455m, 1440m, 1415m, 1405m, 1385m, 1370m, 1360m, 1285w, 1270m, 1250m (sh), 1245s, 1190w, 1160w, 1140w, 1105w, 1070w, 1000w, 930w, 910m, 905w, 890w. <sup>1</sup>H-NMR (300 MHz): 0.18 (*s*, 2 CH<sub>3</sub>Si); 1.00 (*s*, 3 CH<sub>3</sub>CSi); 1.60, 1.68 (*2m*,  $w_{V_2} = 4$ ), 3 H–C(7'), CH<sub>3</sub>–C(6')); 1.82 (*d*, *J* = 1.2, CH<sub>3</sub>–C(2')); 1.90–2.03 (*m*, 2 H–C(4')); 2.03–2.13 (*m*, 2 H–C(3')); 2.42 -2.49, 2.51–2.61, 3.07–3.19 (3m, 2 H–C(4), 2 H–C(5), 2 H–C(6)); 5.13 (*tm*, *J* = 7,  $w_{V_2} = 4$ , H–C(5')); 5.60 (*m*,  $w_{V_3} = 4$ , H–C(1')). <sup>13</sup>C-NMR (75 MHz; *ca*. 80% pure): -6.4 (*q*, 2 CH<sub>3</sub>Si); 17.8, 25.2<sup>17</sup>) (3*q*, CH<sub>3</sub>–C(2'), CH<sub>3</sub>–C(6')); 128.5 (*q*, 3 CH<sub>3</sub>CSi); 25.9<sup>17</sup>), 26.4, 30.6 (5*t*, C(4), C(5), C(6), C(3')), C(4')); 124.8, 128.5 (2*d*, C(1'), C(5')); 19.8 (*s*, CSi); 45.2 (*s*, C(2)); 131.3, 137.4 (2*s*, C(2'), C(6')). MS: 356 (2,  $M^+$ , C<sub>19</sub>H<sub>36</sub>S<sub>2</sub>Si), 221 (12), 167 (16), 135 (10), 133 (12), 125 (11), 115 (13), 91 (11), 73 (100), 69 (49), 59 (13), 41 (39). Anal. calc. for C<sub>19</sub>H<sub>36</sub>S<sub>2</sub>Si (356.71): C 63.98, H 10.17; found: C 64.23, H 10.20.

1.1.3. Dethioacetalization of (E/Z)-14. 1.1.3.1. With  $Tl(NO_3)_3 \cdot 3 H_2O$ . To a soln. of (E/Z)-14 (2.50 g, 7.01 mmol; 3:2 mixture) in THF (50 ml) and H<sub>2</sub>O (70 drops) was added rapidly, at 0°, a soln. of  $Tl(NO_3)_3 \cdot 3 H_2O$  (3.00 g, 7.68 mmol) in MeOH (*Fluka*, 40 ml). After stirring for 5 min at r.t., the mixture was diluted with hexane, filtered through *Celite*, washed with aq. sat. NaCl, and dried (MgSO<sub>4</sub>). CC (pentane/Et<sub>2</sub>O 25:1) afforded (*E*)-7 (608 mg, 33%), and (*Z*)-7 (323 mg, 17%).

(E)-*l*-*[*(tert-*Butyl*)*dimethylsilyl*]-3,7-*dimethyl*-2,6-*octadien*-*l*-*one* ((*E*)-7). B.p. 118–120°/0.07 Torr. UV (0.212 mg in 20 ml): 254 (13830). UV (1.644 mg in 2 ml): 441 (145), 461 (135), 490 (60). IR: 2940s, 2920s, 2880s, 2850s, 1625s, 1560s (br.), 1460m (sh), 1455m, 1440m, 1410m, 1380m, 1360m, 1320w, 1245s, 1190w, 1100w, 1070w, 1000w, 935w. <sup>1</sup>H-NMR (300 MHz, C<sub>6</sub>D<sub>6</sub>): 0.16 (*s*, 2 CH<sub>3</sub>Si): 0.97 (*s*, 3 CH<sub>3</sub>CSi); 1.48, 1.63 (2*m*,  $w_{1/2} = 4$ , CH<sub>3</sub>–C(7), 3 H–C(8)); 1.94–1.96, 2.04–2.08 (2*m*, 2 H–C(4), 2 H–C(5)); 2.12 (*d*, *J* = 1, CH<sub>3</sub>–C(3)); 5.06 (*tm*, *J* = 7.5,  $w_{1/2} = 4$ , H–C(6)); 6.58 (*q*, *J* = 1, H–C(2)). <sup>13</sup>C-NMR (75 MHz, C<sub>6</sub>D<sub>6</sub>): -6.9 (*q*, 2 CH<sub>3</sub>Si); 17.8, 19.6, 25.8 (3*q*, CH<sub>3</sub>–C(3), CH<sub>3</sub>–C(7), C(8)); 26.8 (*q*, 3 CH<sub>3</sub>CSi); 26.4, 41.1 (2*t*, C(4), C(5)); 123.8, 127.9 (2*d*, C(2), C(6)); 16.9 (*s*, CSi); 131.9, 150.9 (2*s*, C(3), C(7)); 234.7 (*s*, C(1)). MS: 266 (1, *M*<sup>++</sup>, C<sub>16</sub>H<sub>30</sub>OSi), 251 (6), 209 (4), 141 (20), 127 (13), 75 (46), 73 (100), 69 (29), 59 (11), 41 (20).

(Z)-1-[(tert-Butyl)dimethylsilyl]-3,7-dimethyl-2,6-octadien-1-one ((Z)-7). B.p. 118–120°/0.07 Torr. UV (0.489 mg in 20 ml): 256 (8200). UV (1.864 mg in 2 ml): 443 (95), 461 (90), 490 (45). IR : 2945s, 2920s, 2895s (sh), 2880s, 2850s, 1685w, 1625s, 1565s (sh), 1560s, 1555s (sh), 1455m, 1440m, 1405w, 1385m, 1370m, 1360m, 1320w, 1290w, 1245s, 1225m, 1155w, 1150w, 1080w, 1000w, 935w. <sup>1</sup>H-NMR (300 MHz, C<sub>6</sub>D<sub>6</sub>): 0.15 (s, 2 CH<sub>3</sub>Si); 0.96 (s, 3 CH<sub>3</sub>CSi); 1.60 (d, J = 1.5, CH<sub>3</sub>–C(3)); 1.63, 1.65 (2m,  $w_{V_2} = 4$ , CH<sub>3</sub>–C(7), 3 H–C(8)); 2.24 (ddd,  $J_1 = J_2 = J_3 = ca. 8, 2 H–C(5)$ ); 2.65 (dd, J = 8, 8, 2 H–C(4)); 5.27 (tm,  $J = 7.5, w_{V_2} = 4$ , H–C(6)); 6.54 (m,  $w_{V_2} = 4$ , H–C(2)). <sup>13</sup>C-NMR (75 MHz, C<sub>6</sub>D<sub>6</sub>): -6.8 (q, 2 CH<sub>3</sub>Si); 17.8, 25.4, 25.9 (3q, CH<sub>3</sub>–C(3), CH<sub>3</sub>–C(7), C(8)); 26.8 (q, 3 CH<sub>3</sub>CSi); 27.7, 34.7 (2t, C(4), C(5)); 124.5, 128.7 (2d, C(2), C(6)); 17.0 (s, CSi); 131.8, 152.4 (2s, C(3), C(7)); 234.7 (s, C(1)). MS. 266 (1,  $M^{++}$ , C<sub>16</sub>H<sub>30</sub>OSi), 251 (8), 209 (4), 141 (12), 115 (17), 75 (76), 73 (100), 69 (28), 59 (11), 41 (20).

1.1.3.2. With  $HgCl_2/HgO$ . To a soln. of (E/Z)-14 (2.52 g, 7.06 mmol; ca. 2:1 mixture) in MeOH (*Fluka*, 38 ml) HgCl<sub>2</sub> (5.03 g, 18.5 mmol) and HgO (2.23 g, 10.3 mmol) was added. The mixture was heated at reflux temp. for 30 min, filtered through *Celite*, and worked up in Et<sub>2</sub>O. CC (hexane/Et<sub>2</sub>O 30:1) afforded (E/Z)-7 (1.06 g, 56%).

1.2. Acylsilane (E)-8. 1.2.1. Transformation of 19 to 21. To a soln. of  $I_2$  (28 g, 110 mmol) in pyridine (42 ml) was added 19 (21 g, 102 mmol) at r.t. The mixture was stirred at 100° for I h, concentrated under reduced pressure, stirred with 2N NAOH (330 ml) at 100° overnight, cooled to 0°, acidified with 10% aq. HCl, and extracted with Et<sub>2</sub>O. The org. phase was extracted with 2N aq. Na<sub>2</sub>CO<sub>3</sub>, and the aq. phase was acidified again with 10% HCl and extracted with  $Et_2O$ . After washing with 20% aq. Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>, this org. phase was worked up affording (E)-2-methyl-3-(2,5,5-trimethyl-2-cyclohexenyl)acrilic acid (15.8 g, 74%). This material was dissolved in abs. Et<sub>2</sub>O (100 ml) and added dropwise over 10 min to a suspension of LiAlH<sub>4</sub> (3.67 g, 96.6 mmol) in abs. Et<sub>2</sub>O (450 ml). After stirring for 30 min at r.t., the mixture was worked up by adding *Celite*, sat. aq. NH<sub>4</sub>Cl, and MgSO<sub>4</sub>. Distillation (80°/0.07 Torr) afforded 21 (9.38 g, 65%).

1.2.2. Oxidation of **21** to **11**. A soln. of **11** (9.38 g, 48.4 mmol) in hexane (900 ml) was stirred with MnO<sub>2</sub> (61 g, 70.1 mmol) overnight under Ar. The mixture was filtered through *Celite* and evaporated under reduced pressure affording crude **11** (8.1 g, 87%).

<sup>&</sup>lt;sup>17</sup>) Two signals overlapping.

1.2.3. Transformation of 11 into 15. A soln. of 1,3-propanedithiol (5 ml, 49.8 mmol) and  $BF_3 \cdot OEt_2$  (3.7 ml, 29.5 mmol) in AcOH (24 ml, Merck) was added dropwise to a soln. of 11 (8.1 g, 42.2 mmol) in AcOH (Merck, 80 ml) and abs.  $CH_2Cl_2$  (80 ml) at 0°. The mixture was stirred for 45 min at 0° and for 3 h at r.t. After the addition of ice (150 g), it was diluted with  $Et_2O$ , washed with 2N NaOH (twice), 0.5N NaHCO<sub>3</sub> (twice), and sat. NaCl, and dried (MgSO<sub>4</sub>). CC (SiO<sub>2</sub>, hexane/Et<sub>2</sub>O 50:1) gave 15 (8.9 g, 75%).

2-[1'-Methyl-2'-(2",6",6",6"-trimethyl-2"-cyclohexenyl)vinyl]-1,3-dithiane (15). IR: 3010w (sh), 2950s, 2890s, 2850s (sh), 1440m, 1425m, 1415m, 1375m, 1355m, 1270m, 1170m, 1115w (br.), 1070w, 1005w (br.), 960w, 910w. <sup>1</sup>H-NMR (80 MHz, CDCl<sub>3</sub>): 0.83, 0.90 (2s, 2 CH<sub>3</sub>-C(6")); 1.08-1.75 (m, 2 H-C(5), H-C(5")); 1.55 (m,  $w_{1/2} = 4$ , CH<sub>3</sub>-C(1')); 1.80-2.18 (m, 2 H-C(4"), H-C(5")); 1.88 (s, CH<sub>3</sub>-C(2')); 2.45 (d, J = 11, H-C(1")); 2.75-3.05 (m, 2 H-C(4), 2 H-C(6)); 4.58 (s, H-C(2)); 5.25-5.43 (m,  $w_{1/2} = 8$ , H-C(3")); 5.43 (d, J = 11, H-C(2')). MS: 282 (73,  $M^{++}$ , C<sub>16</sub>H<sub>26</sub>S<sub>2</sub>), 226 (10), 176 (10), 175 (24), 161 (20), 159 (14), 152 (32), 151 (43), 137 (46), 125 (12), 121 (10), 120 (47), 119 (100), 111 (11), 107 (23), 106 (38), 105 (29), 91 (21), 79 (12), 77 (13), 55 (12), 53 (11), 45 (11), 41 (29).

1.2.4. Transformation of 15 into 16. To a soln. of 15 (8.9 g, 31.6 mmol) in abs. THF (150 ml) and abs. HMPA (5.9 ml) was added under Ar at  $-78^{\circ}$  BuLi (1.6M in hexane, 24 ml, 38.4 mmol). The mixture was stirred for 1 h, a soln. of (*t*-Bu)Me<sub>2</sub>SiCl (5.6 g, 37.2 mmol) in abs. THF (30 ml) was added dropwise, and the mixture was allowed to warm to r.t. slowly. After stirring for 1 h, the mixture was worked up with Et<sub>2</sub>O, and CC (hexane:Et<sub>2</sub>O 50:1) gave 16 (9.5 g, 76%).

(E)-2-[(tert-Butyl)dimethylsilyl]-2-[1'-methyl-2'-(2", 6", 6"-trimethyl-2"-cyclohexenyl)vinyl]-1,3-dithiane (16). M.p. 50-53". IR: 3020w (sh), 2940s, 2920s, 2890s, 2850s, 2800w (sh), 1430m (sh), 1455m, 1415m, 1405m, 1385m, 1370m, 1355m, 1340w, 1290w, 1265m, 1250s (sh), 1245s, 1120w (br.), 1000m, 915m, 880m, 830s. <sup>1</sup>H-NMR (80 MHz, CDCl<sub>3</sub>): 0.15 (s, 2 CH<sub>3</sub>Si); 0.90, 0.95 (2s, 2 CH<sub>3</sub>-C(6")); 1.06 (s, 3 CH<sub>3</sub>CSi); 1.20-1.78 (m, 2 H-C(5), H-C(5")); 1.63 (m,  $w_{V_2} = 4$ , CH<sub>3</sub>-C(2")); 1.80-2.05 (m, 2 H-C(4"), H-C(5")); 1.98 (s, CH<sub>3</sub>-C(1')); 2.25-3.18 (m, 2 H-C(4), 2 H-C(6), H-C(1")); 5.22-5.42 (m,  $w_{V_2} = 8$ , H-C(3")); 5.84 (d, J = 11, H-C(2')). MS: 396 (18,  $M^{+*}$ , C<sub>22</sub>H<sub>40</sub>S<sub>2</sub>Si), 340 (3), 339 (4), 281 (40), 159 (18), 151 (19), 149 (42), 147 (11), 137 (17), 129 (14), 127 (11), 123 (11), 119 (19), 115 (14), 111 (11), 109 (11), 107 (14), 97 (21), 95 (18), 93 (10), 91 (13), 85 (18), 84 (12), 83 (24), 82 (12), 81 (43), 73 (100), 71 (25), 70 (15), 69 (94), 60 (25), 59 (11), 57 (44), 55 (40), 43 (41), 41 (46). Anal. calc. for C<sub>22</sub>H<sub>40</sub>S<sub>2</sub>Si (396.78): C 60.60, H 10.16, S 16.16; found: C 66.68, H 10.34, S 15.91.

1.2.5. Dethioacetalization of 16. 1.2.5.1. With  $Tl(NO_3)_3 \cdot 3 H_2O$ . To a soln. of 16 (9.5 g, 24 mmol) in THF (150 ml) and H<sub>2</sub>O (50 drops) was added at 0° at once a soln. of  $Tl(NO_3)_3 \cdot 3 H_2O$  (12 g, 27 mmol) in abs. MeOH (*Fluka*, 180 ml). After stirring at r.t. for 5 min, the mixture was diluted with hexane, filtered through Celite, washed with sat. aq. NaCl, and dried (MgSO<sub>4</sub>). CC (pentane/Et<sub>2</sub>O 50:1) yielded (E)-8 (3.3 g, 45%).

 $\begin{array}{l} (E)^{-1/[(\text{tert-}Butyl)dimethylsilyl]-2-methyl-3-(2', 6', 6'-trimethyl-2'-cyclohexenyl)-2-propen-1-one} \\ ((E)-8). \\ \text{M.p. 56-57}^\circ. B.p. 60^\circ/0.07 \text{ Torr. UV} (0.134 \text{ mg in 10 ml}): 244 (13960). UV (1.02 \text{ mg in 2 ml}): 418 (130). IR: 3020w, \\ 2950s, 2920s, 2850s, 1630w (br.), 1575s, 1465m (sh), 1375m, 1360m, 1275m, 1245m, 1225w, 1205w, 1170m, 1025w, \\ 1005w (sh), 960w, 945w (br.), 895w, 835m. ^{1}\text{H-NMR} (300 \text{ MHz, } C_{5}D_{6}): 0.24, 0.25 (2s, 2 \text{ CH}_{3}\text{S}); 0.83, 0.88 (2s, 2 \text{ CH}_{3}-C(6')); 0.95 (s, 3 \text{ CH}_{3}\text{CS}); 1.13-1.20 (m, H-C(5')); 1.45-1.53 (m, H-C(5')); 1.50 (m, w_{1/2} = 4, \text{ CH}_{3}-C(2')); \\ 1.82 (d, J = 1.5, \text{ CH}_{3}-C(2)); 1.95-1.99 (m, w_{1/2} = 15, 2 \text{ H}-C(4')); 2.65 (d, J = 11, \text{ H}-C(1')); 5.37 (m, w_{1/2} = 8, \text{ H}-C(3')); 6.43 (dd, J = 11, 15, \text{ H}-C(3)). ^{13}\text{C-NMR} (75 \text{ MHz, } C_{6}D_{6}): - 3.9, - 3.5 (2q, 2 \text{ CH}_{3}\text{S}); 10.8 (q, \text{ CH}_{3}-C(2)); 23.0 (q, \text{ CH}_{3}-C(2')); 150.0 (d, C(3)); 17.2 (s, \text{ CS}); 33.0 (s, C(4')); 23.4 (t, C(5')); 32.2 (t, C(4')); 50.7 (d, C(1')); 122.0 (d, C(3')); 150.0 (d, C(3)); 17.2 (s, \text{ CS}); 33.3 (s, C(1)). \text{ MS: 306 } (1, M^{++}, \text{ C}_{19}\text{ H}_{34}\text{ OSi}), 250 (13), 183 (46), 179 (19), 127 (100), 119 (13), 107 (10), 105 (10), 91 (22), 75 (25), 73 (94), 59 (12). \text{ Anal. calc. for C}_{19}\text{ H}_{34}\text{ OSi} (306.38): C 74.44, H 11.18; found: C 74.47, H 11.30. \end{array}$ 

1.2.5.2. With  $HgCl_2/HgO$ . To a soln. of 16 (191 mg, 0.48 mmol) in MeOH (3.5 mol) was added HgCl\_2 (452 mg, 1.66 mmol) and HgO (201 mg, 0.93 mmol). The mixture was heated under reflux for 30 min, filtered through *Celite*, and worked up in Et<sub>2</sub>O. CC (hexane/Et<sub>2</sub>O 10:1) afforded (*E*)-8 (137 mg, 93%).

1.3. Acylsilane (E)-9. 1.3.1. Transformation of 20 into 17. The reaction sequences analogous to those described in Sect. 1.2.1 to 1.2.3 gave 17.

(E)-2-[2'-(2",6",6",6"-Trimethyl-2"-cyclohexenyl)vinyl]-1,3-dithiane (17). B.p. 150°/0.05 Torr. IR: 3030m, 2950s, 2900s, 2850s, 2810w (sh), 1465m, 1445m, 1430m, 1420s, 1415m, 1385m, 1375m, 1365m, 1345w, 1300w, 1275s, 1240w, 1175m (sh), 1170m, 1135w, 1130w, 1075w, 975m, 965m, 960m (sh), 930w (br.), 910m, 880w, 865w. <sup>1</sup>H-NMR (100 MHz, CCl<sub>4</sub>): 0.83, 0.88 (2s, 2 CH<sub>3</sub>-C(6")); 1.00-1.52 (m, 2 H-C(5), H-C(5")); 1.58 (m,  $w_{Y_4} = 4$ , CH<sub>3</sub>-C(2")); 1.84-2.16 (m, 2 H-C(4"), H-C(5"), H-C(1")); 2.68-2.88 (m, 2 H-C(4), 2 H-C(6)); 4.50 (d, J = 7, H-C(2)); 5.35 (m,  $w_{Y_4} = 8$ , H-C(3")); 5.51 (*AB*, J = 16,  $\delta_A = 5.40$ ,  $\delta_B = 5.62$ , split into d, J = 8, H-C(1'), H-C(2')). MS: 268 (100,  $M^+$ , C<sub>15</sub>H<sub>24</sub>S<sub>2</sub>), 212 (8), 197 (9), 161 (19), 147 (11), 145 (17), 138 (21), 137 (17), 132 (11), 123 (33), 119 (19), 107 (14), 106 (33), 105 (25), 91 (22), 79 (14), 77 (14), 41 (23). Anal. calc. for C<sub>15</sub>H<sub>24</sub>S<sub>2</sub> (268.49): C 67.10, H 9.00, S 23.89; found: C 66.96, H 8.74, S 24.16.

1.3.2. Transformation of 17 into 18. The reaction of a soln. of 10 (7.0 g, 26.1 mmol) in abs. THF (150 ml) and HMPA (4.7 ml) with BuLi (1.6m in hexane; 19 ml, 30.4 mmol) and (t-Bu)Me<sub>2</sub>SiCl (4.43 g, 29.4 mmol) in abs. THF (20 ml) as described in Sect. 1.2.4 yielded 18 (7.34 g, 73%).

(E)-2-[(tert-Butyl)dimethylsilyl]-2-[2'-(2",6",6",6"-trimethyl-2"-cyclohexenyl)vinyl]-1,3-dithiane (18). B.p. 180°/0.05 Torr. 1R: 3010w, 2960s, 2930s, 2850s, 1465m, 1460m, 1445m, 1430m, 1420m, 1410m, 1390m, 1380m, 1375w, 1345w, 1270m, 1255m, 1245s, 1180w, 1140w, 1075w, 1045w, 1025w, 1005w, 980m, 940w (sh), 930w, 920m, 890w, 865w, 830s. <sup>1</sup>H-NMR (100 MHz, CCl<sub>4</sub>): 0.09 (s, 2 CH<sub>3</sub>Si); 0.92, 0.94 (2s, 2 CH<sub>3</sub>-C(6")); 1.02 (s, 3 CH<sub>3</sub>CSi); 1.12–1.60 (m, 2 H-C(5), H-C(5")); 1.66 (m,  $w_{1/2} = 4$ , CH<sub>3</sub>-C(2")); 1.86–2.44 (6 H), 2.74–3.12 (2 H) (2m, 2 H-C(4), 2 H-C(6), 2 H-C(4"), H-C(1"), H-C(5")); 5.37 (m,  $w_{1/2} = 8$ , H-C(3")); 5.57 (*AB*, *J* = 15,  $\delta_A = 5.52$ ,  $\delta_B = 5.62$ , split into *d*, *J* = 7, H-C(2'), H-C(1')). MS: 382 (24,  $M^{++}$ , C<sub>21</sub>H<sub>38</sub>S<sub>2</sub>Si), 267 (38), 259 (48), 235 (28), 211 (17), 179 (14), 161 (12), 145 (69), 137 (17), 119 (14), 115 (19), 113 (48), 105 (22), 91 (28), 73 (100), 59 (22), 41 (20). Anal. calc. for C<sub>21</sub>H<sub>38</sub>S<sub>2</sub>Si (382.75): C 65.90, H 10.01, S 16.75; found: C 66.01, H 10.02, S 16.60.

1.3.3. Transformation of **18** into (E)-9. The reaction of **18** (7.34 g, 19.2 mmol) in THF (120 ml) with  $Tl(NO_{3})_{3} \cdot 3 H_{2}O$  (9.65 g, 21.7 mmol) in MeOH (150 ml) as described in Sect. 1.2.5 gave (E)-9 (5.0 g, 67%).

(E)-*1*-[(tert-*Butyl*)*dimethylsilyl*]-*3*-(2',6',6'-trimethyl-2'-cyclohexenyl)-2-propen-1-one ((E)-9). B.p. 55°/0.07 Torr. UV: (0.18 mg in 10 ml) 225 (12990), (1.195 mg in 2 ml) 412 (120). IR: 3020w, 2950s, 2920s, 2880m (sh), 2850s, 1635m, 1620m (sh), 1580s, 1555m (sh), 1465m (sh), 1455m, 1430w, 1380w, 1360m, 1285w (br.), 1245s, 1185w, 1155m, 980m, 935w. <sup>1</sup>H-NMR (300 MHz, C<sub>6</sub>D<sub>6</sub>): 0.19 (*s*, 2 CH<sub>3</sub>Si); 0.81, 0.83 (2*s*, 2 CH<sub>3</sub>-C(6')); 0.96 (*s*, 3 CH<sub>3</sub>CSi); 1.00-1.10, 1.30-1.42 (2*m*, 2 H-C(5')); 1.52 (*m*,  $w_{V_2} = 5$ , CH<sub>3</sub>-C(2')); 1.89 (*m*,  $w_{V_2} = 15$ , 2 H-C(4')); 2.08 (*d*, *J* = 9.5, H-C(1')); 5.36 (*m*,  $w_{V_2} = 8$ , H-C(3')); 6.40 (*d*, *J* = 15.5, H-C(2')); 6.60 (*d*, *J* = 15.5, 9.5, H-C(3)). <sup>13</sup>C-NMR (75 MHz, C<sub>6</sub>D<sub>6</sub>): - 5.58 (2*q*, 2 CH<sub>3</sub>Si); 23.0 (*q*, CH<sub>3</sub>-C(2')); 1.82 (*d*, 3, 3, (*d*, C(2)); 147.0 (*d*, C(3)); 7.9 (*q*, CH<sub>3</sub>-C(6')); 132.5 (*s*, C(2')); 233.2 (*s*, C(1)). MS: 292 (1, *M*<sup>++</sup>, C<sub>18</sub>H<sub>32</sub>OSi), 250 (10), 183 (32), 179 (13), 128 (11), 127 (93), 119 (11), 91 (15), 75 (28), 73 (100), 59 (12). Anal. calc. for C<sub>18</sub>H<sub>32</sub>OSi (292.54): C 73.90, H 11.03; found: C 73.88, H 11.08.

**2.** Photolyses. – 2.1. Photolyses of (E/Z)-7. 2.1.1. Irradiation ( $\lambda > 347$  nm; lamp B, filter A, 100 % conversion) of (E)-7 (543 mg, 2.04 mmol) in abs. MeCN (200 ml) followed by CC (hexane/Et<sub>2</sub>O 25:1 and 0.1 % of Et<sub>3</sub>N) of the mixture gave: **27A** + **B** (*ca.* 1:1; 120 mg, 22 %), **28** (67 mg, 12 %). **29A** + **B** (*ca.* 1:1; 74 mg, 10 %), and (E/Z)-10 (52 mg, 17%).

2.1.2. Irradiation of a soln. of (Z)-7 (200 mg, 0.75 mmol) in abs. MeCN (200 ml) as described above (conversion 83%) gave after CC (hexane/Et<sub>2</sub>O 40:1 and 0.1% of Et<sub>3</sub>N): 27A + B (*ca.* 1:1; 40 mg, 24%), 28 (33 mg, 19%), 29A + B (*ca.* 1:1; 23 mg, 14%), and (*E*/Z)-10 (15 mg, 16%).

(5 E/Z, 7 E/Z, 11 E/Z) -8- [ (tert-Butyl) dimethylsilyl]-10-[ (tert-butyl) dimethylsilyloxy]-2,6,12,16-tetramethyl-9-oxaheptadeca-2,5,7,11,15-pentaene (**27A** + **B**). B.p. 132°/0.06 Torr. UV (0.406 mg in 20 ml): 233 (sh) (7600). IR: 3020w (sh), 2950s, 2930s, 2890s, 2880s, 2850s, 2730w, 2720w, 1665w, 1610w, 1585w, 1470m, 1460m, 1445m, 1410w, 1390w, 1375m, 1360w, 1250s, 1155w, 1140m (br.), 1105 m, 1095m, 1065m (sh), 1040m, 1025m, 1005m, 965m (br.), 940m, 870w (sh). <sup>1</sup>H-NMR (300 MHz): 0.006, 0.01, 0.052, 0.057, 0.06, 0.071, 0.074, 0.078 (8s, CH<sub>3</sub>Si); 0.86, 0.87, 0.88, 0.91 (4s, 3 CH<sub>3</sub>CSi); 1.52, 1.60, 1.63, 1.67, 1.68, 1.69, 1.71, 1.72, 1.76, 1.77 (10s, 3 H–C(1), CH<sub>3</sub>–C(2), CH<sub>3</sub>–C(6), CH<sub>3</sub>–C(12), CH<sub>3</sub>–C(16), 3 H–C(17)); 1.95–2.20 (m, 2 H–C(13), 2 H–C(14)); 2.70 (m, J = 6.5,  $w_{\gamma_2} = 4$ , 2 H–C(4)); 5.05–5.25 (m, H–C(3), H–C(5), H–C(15)); 5.35, 5.38 (2dm, overlapping, J = 7,  $w_{\gamma_2} = 4$ , H–C(11)); 5.77, 5.80 (2d, overlapping, J = 7, H–C(10)); 6.18 (m,  $w_{\gamma_2} = 5$ , H–C(7)). <sup>13</sup>C-NMR (75 MHz): -4.0, -3.3 (2q, CH<sub>3</sub>Si); 17.0, 17.7, 17.8, 23.1, 25.8, 26.6, 27.0 (7q, C(1), CH<sub>3</sub>–C(2), CH<sub>3</sub>–C(6), CH<sub>3</sub>–C(10), CH<sub>3</sub>–C(12), CH<sub>3</sub>–C(16), C(17)); 25.9, 27.6 (2q, 3 CH<sub>3</sub>CSi); 26.2, 32.6, 39.3 (3t, C(4), C(13), C(14)); 93.4, 93.9 (2d, C(10)); 124.0, 124.1, 124.2, 126.4, 126.6, 126.9 (6d, C(3), C(5), C(7), C(11), C(15)); 17.4, 18.1 (2s, CSi); 131.4, 131.8, 138.0 (3s, C(2), C(6), C(12), C(16)); 158.2, 158.3 (2s, C(8)). MS: S32 ( < 1, M<sup>++</sup>, C<sub>32</sub>H<sub>60</sub>O<sub>2</sub>Si<sub>2</sub>), 517 ( < 1), 475 ( < 1), 401 ( < 1), 343 (1), 268 (13), 267 (57), 173 (36), 135 (32), 115 (11), 107 (17), 93 (14), 75 (34), 73 (100), 69 (50), 59 (15), 41 (28).

(E)-*I*-*[*(tert-*Butyl*)*dimethylsily1*-3,7-*dimethyl*-3,6-octadien-1-one (**28**). B.p. 100°/0.08 Torr. UV (2.118 mg in 2 ml): 346 (80), 361 (115), 374 (150), 388 (150). IR: 3020w, 2950s, 2920s, 2880s, 2850s, 1715w, 1640m, 1625s, 1465m, 1460m, 1440m (sh), 1390m (sh), 1380m, 1375m, 1360m, 1295w, 1270w, 1245s, 1180w (br.), 1000w, 935w. <sup>1</sup>H-NMR: 0.2 (s, 2 CH<sub>3</sub>Si); 0.96 (s, 3 CH<sub>3</sub>CSi); 1.61, 1.65, 1.71 (3m,  $w_{1/2} = 3$ , CH<sub>3</sub>-C(3), CH<sub>3</sub>-C(7), 3 H-C(8)); 2.74 (*tm*,  $J = 6.5, w_{1/2} = 4, 2$  H-C(5)); 3.27 (s, 2 H-C(2)); 5.00-5.30 (m, H-C(6), H-C(4)). <sup>13</sup>C-NMR: -6.5 (g, 2 CH<sub>3</sub>Si); 168, 17.7, 25.6 (3g, CH<sub>3</sub>-C(3), CH<sub>3</sub>-C(7), C(8)); 26.4 (g, 3 CH<sub>3</sub>CSi); 27.2 (*t*, C(5)); 60.5 (*t*, C(2)); 122.6, 128.7 (2d, C(4), C(6)); 16.6 (s, CSi); 127.8, 131.7 (2s, C(3), C(7)); ca. 240 (s, C(1)). MS: 266 (< 1,  $M^{++}$ , C<sub>16</sub>H<sub>30</sub>OSi), 251 (1), 238 (1), 209 (1), 143 (8), 115 (52), 75 (10), 73 (100), 59 (8), 41 (7).

(tert-Butyl)dimethylsilyl [1-Methyl-2-(2'-methyl-1'-propenyl)cyclopropyl]methyl Ketone, Isomer A (29A). UV (1.479 mg in 2 ml): 349 (sh) (60), 364 (100), 378 (140), 395 (115). IR: 3060w, 3010m (sh), 2950s, 2930s, 2880s, 2860s, 2740w, 2730w, 2720w, 1710w, 1690w, 1645s, 1470s, 1460s, 1445s, 1400m (sh), 1390m, 1375s, 1360m, 1335m, 1325m (sh), 1275w, 1255s (sh), 1250s, 1170w, 1145w, 1075w, 1030m, 1005m, 985w, 940m. <sup>1</sup>H-NMR (80 MHz): 0.10–0.40, 0.70–0.90 (2m, 2 H–C(3)); 0.20 (s, 2 CH<sub>3</sub>Si); 0.96 (s, 3 CH<sub>3</sub>CSi); 1.05–1.40 (m, H–C(2)); 1.14 (s, CH<sub>3</sub>–C(1)); 1.70, 1.73 (2m,  $w_{Y_6}$  = 3, CH<sub>3</sub>–C(2'), 3 H–C(3')); 2.61 (*AB*, *J* = 18,  $\delta_A$  = 2.49,  $\delta_B$  = 2.74, CH<sub>2</sub>CO); 4.69 (dm, *J* = 7,  $w_{Y_6}$  = 4, H–C(1')). <sup>13</sup>C-NMR (75 MHz): -6.9, -6.8 (2q, 2 CH<sub>3</sub>Si); 18.4, 25.3, 25.8 (3q, CH<sub>3</sub>–C(1), CH<sub>3</sub>–C(2'), C(3')); 26.5 (q, 3 CH<sub>3</sub>CSi); 21.8 (t, C(3)); 55.2 (t, CH<sub>2</sub>CO); 22.8 (d, C(2)); 124.4 (d, C(1')); 16.7, 18.8 (2s, CSi, C(1)); 133.9 (s, C(2')); 246.8 (s, CO). MS: 266 ( < 1,  $M^+$ , C<sub>16</sub>H<sub>30</sub>OSi), 251 (1), 225 ( < 1), 209 (1), 143 (3), 115 (36), 75 (10), 73 (100), 41 (7).

*Isomer B* (29B). UV (2.174 mg in 2 ml): 350 (sh) (60), 366 (100), 381 (140), 398 (128). IR: 3060w, 3010w (sh), 2960s, 2930s, 2900s, 2880s, 2860s, 2740w, 2730w, 2720w, 1700w (br.), 1640s, 1470m, 1460s, 1445m, 1410w, 1390m, 1380m, 1365m, 1335w, 1320w, 1300w (sh), 1255m (sh), 1250s, 1170w, 1140w, 1080w, 1035w, 1025w, 1010w, 990w, 940m. <sup>1</sup>H-NMR (300 MHz): 0.16 (*s*, 2 CH<sub>3</sub>Si); 0.28 (*dd*,  $J_1 = J_2 = 5.2$ ) and 0.70 (*dd*, J = 8.6, 4.8, 2 H–C(3)); 0.92 (*s*, 3 CH<sub>3</sub>CSi); 1.02 (*s*, CH<sub>3</sub>–C(1)); 1.16–1.50 (*m*, H–C(2)); 1.71 (*s*, CH<sub>3</sub>–C(2'), 3 H–C(3')); 2.58 (*AB*, J = 17,  $\delta_A = 2.52$ ,  $\delta_B = 2.64$ , CH<sub>2</sub>CO); 4.85 (*dm*, J = 8.1,  $w_{12} = 4$ , H–C(1')). <sup>13</sup>C-NMR (75 MHz): -6.9 (*q*, 2 CH<sub>3</sub>Si); 184, 19.2, 25.9 (3*q*, CH<sub>3</sub>–C(1), CH<sub>3</sub>–C(2'), C(3')); 26.6 (*q*, 3 CH<sub>3</sub>CSi); 21.1 (*t*, C(3)); 60.9 (*t*, CH<sub>2</sub>CO); 22.3 (*d*, C(2)); 123.9 (*d*, C(1')); 16.8, 18.0 (2*s*, CSi, C(1)); 133.3 (*s*, C(2')); 247.2 (*s*, CO). MS: 266 ( < 1,  $M^+$ , C<sub>16</sub>H<sub>30</sub>OSi), 251 ( < 1), 209 ( < 1), 143 (4), 115 (45), 75 (6), 73 (100), 59 (8), 41 (7).

2.1.3. Photolysis of (E/Z)-7 in the Presence of  $(t-Bu)Me_2SiOH$ . A soln. of (E/Z)-7 (ca. 2:1; 326 mg, 1.22 mmol) and  $(t-Bu)Me_2SiOH$  (262 mg, 1.98 mmol) in abs. MeCN (100 ml) was irradiated as described above (95% conversion). CC (hexane/Et<sub>2</sub>O 50:1) afforded: (E/Z)-45 (110 mg, 24%), 28 (40 mg, 13%), 29A + B (ca. 1:1; 25 mg, 8%), and (E/Z)-10 (43 mg, 24%).

(E/Z)-3,7-Dimethyl-2,6-octadien-1-al-bis[ (tert-butyl) dimethylsilyl]-acetal ((E/Z)-45). IR: 2950s, 2930s, 2890s, 2850s, 2740w, 2710w, 1665w, 1640w, 1470m, 1460s, 1440m, 1405w, 1390m, 1380m, 1360m, 1250s, 1215w, 1185w, 1150m (sh), 1130s, 1105m, 1070s, 1025s (br.), 1000s, 940m, 890m. <sup>1</sup>H-NMR (80 MHz): 0.13 (s, 2 CH<sub>3</sub>Si); 0.91 (s, 3 CH<sub>3</sub>CSi); 1.65, 1.73 (2m,  $w_{Y_4}$  = 4, CH<sub>3</sub>-C(3), CH<sub>3</sub>-C(7), 3 H-C(8)); 1.95-2.25 (m, 2 H-C(4), 2 H-C(5)); 4.95-5.45 (m, H-C(2), H-C(6)); 5.75, 5.78 (2d, J = 7, H-C(1)). MS: 341 (8,  $M^{++}$  - C<sub>4</sub>H<sub>9</sub>), 267 (6), 263 (7), 247 (10), 221 (13), 182 (18), 181 (100), 147 (20), 135 (23), 125 (26), 121 (25), 113 (26), 107 (21), 99 (10), 97 (13), 93 (11), 75 (17), 74 (14), 73 (98), 69 (32), 59 (38), 41 (23).

2.2. Photolyses of (E)-8. 2.2.1. In THF. A soln. of (E)-8 (522 mg, 1.7 mmol) in abs. THF (140 ml) was irradiated ( $\lambda > 347$  nm, lamp B, filter A, ca. 100% conversion) under Ar. Evaporation of the solvent and CC (SiO<sub>2</sub>, hexane/Et<sub>2</sub>O gradient, O $\rightarrow$ 2.5% Et<sub>2</sub>O) of the residue gave (Z)-8 (115 mg, 22%), **30A** (30 mg, 6%), and **30B** (23 mg, 4%).

2.2.2. In MeCN. A soln. of (E)-8 (460 mg, 1.5 mmol) in abs. MeCN (130 ml) was irradiated as described in Sect. 2.1.1 (98% conversion). CC of the mixture afforded (Z)-8 (97 mg, 21%), 30A (20 mg, 4%), and 30B (3%).

(Z)-1-[ (tert-Butyl) dimethylsilyl]-2-methyl-3-(2',6',6'-trimethyl-2'-cyclohexenyl)-2-propen-1-one ((Z)-8). UV (0.132 mg in 10 ml): 263 (5100). UV (1.243 mg in 2 ml): 402 (130), 418 (130). IR: 3020w (sh), 2950s, 2920s, 2850s, 1625m, 1595m, 1460m (sh), 1455s, 1445s, 1430s, 1405w, 1380m, 1370m, 1360m, 1300w, 1245s, 1195w, 1185w, 1130w, 1080w, 1035w (br.), 1000w (sh), 980m, 935w, 900w (sh), 870w, 830s. <sup>1</sup>H-NMR (80 MHz, C<sub>6</sub>D<sub>6</sub>): 0.25 (s, 2 CH<sub>3</sub>Si); 1.04 (3 H), 1.08 (12 H) (2s, 3 CH<sub>3</sub>CSi, 2 CH<sub>3</sub>-C(6')); 1.15-1.72 (m, 2 H-C(5')); 1.81 (d, J = 1.5, CH<sub>3</sub>-C(2)); 1.90 (m,  $w_{1/2} = 4$ , CH<sub>3</sub>-C(2')); 2.25 (m,  $w_{1/2} = 16$ , 2 H-C(4')); 2.90 (d, J = 11, H-C(1')); 5.15 (dm, J = 11,  $w_{1/2} = 4$ , H-C(3)); 5.44 (m,  $w_{1/2} = 6$ , H-C(3')). MS: 306 (21,  $M^{++}$ , C<sub>19</sub>H<sub>34</sub>OSi), 291 (27), 235 (18), 193 (16), 179 (41), 175 (33), 173 (16), 159 (38), 127 (11), 119 (31), 115 (12), 105 (12), 91 (19), 75 (90), 74 (15), 73 (100), 59 (19), 41 (14).

5-f (tert-Butyl)dimethylsilyl]-3-f (tert-butyl)dimethylsilyloxy]-2.6-dimethyl-1-(2', 6', 6'-trimethyl-2'-cyclohexenyliden)-4-oxa-I,5-heptadiene, Isomer A (**30A**). UV (0.134 mg in 10 ml): 205 (31 650), 245 (14 680). IR: 3020w, 2950s, 2920s, 2850s, 1465m (sh), 1455m. 1380m, 1370m, 1355m, 1325m, 1245s, 1220w (br.), 1170w, 1100m, 1075m, 1030m (sh), 1020s, 1000m (sh), 965m, 935w, 890s, 870m, 830s. <sup>1</sup>H-NMR (300 MHz, C<sub>6</sub>D<sub>6</sub>): 0.20, 0.23, 0.39, 0.50 (4x, 4 CH<sub>3</sub>Si); 0.97, 1.00 (2x, 2 CH<sub>3</sub>-C(6')); 1.05, 1.18 (2s, 6 CH<sub>3</sub>CSi); 1.10-1.26, 1.47-1.60 (2m, 2 H-C(5')); 1.34, 1.43 (2s, 2 CH<sub>3</sub>-C(6')); 1.95, 1.97 (2d,  $J \approx 1$ , CH<sub>3</sub>-C(2), CH<sub>3</sub>-C(6)); 2.03-2.16 (m, 2 H-C(4')); 2.06 (m,  $w_{1/2} = 5$ , CH<sub>3</sub>-C(2'')); 2.49 (d, J = 9, H-C(1')); 5.30 (m,  $w_{1/2} = 9$ , H-C(3'')); 5.96 (s, H-C(3)); 5.96 (s, H-C(3)); 6.07 (s, H-C(7)). <sup>13</sup>C-NMR (75 MHz, C<sub>6</sub>D<sub>6</sub>): -3.7, -3.3, -2.8, -2.6 (4q, 4 CH<sub>3</sub>Si); 11.6 (q, CH<sub>3</sub>-C(2)); 21.9, 22.9, 23.8, 27.3, 27.4, 29.2, 29.5 (7q, CH<sub>3</sub>-C(C2'), 2 CH<sub>3</sub>-C(6')), 2.07, 2 (d, C(1')); 40.1 (t. C(4'')); 49.2 (d, C(1')); 101.2 (d, C(3)); 121.1 (d, C(3')); 125.3 (d, C(3'')); 128.0, 129.5 (2d, C(1), C(7)); 18.9, 19.7 (2s, 2 CSi); 32.9, 35.7 (2s, C(6'), C(6'')); 133.4, 135.9, 137.0,

137.6, 143.3 (5*s*, 1 *s* is hidden or overlapped by *d* at 121.1, 125.3 or 128.0, C(2), C(5), C(6), C(2'), C(1"), C(2")). MS: 612 (1, *M*<sup>++</sup>, C<sub>38</sub>H<sub>68</sub>O<sub>2</sub>Si<sub>2</sub>), 535 (1), 481 (3), 309 (15), 308 (57), *307* (100), 251 (22), 186 (11), 185 (70), 175 (47), 147 (11), 119 (28), 115 (18), 105 (18), 75 (41), 74 (27), 73 (99), 69 (11), 59 (20). Anal. calc. for C<sub>38</sub>H<sub>68</sub>O<sub>2</sub>Si<sub>2</sub> (613.14): C 74.44, H 11.18; found: C 74.36, H 11.24.

*Isomer B* (**30B**), 80% pure. UV (0.122 mg in 10 ml): 206 (24 100), 244 (13 600). IR : 3010*w*, 2950*s*, 2920*s*, 2850*s*, 1465*m* (sh), 1455*m*. 1445*m* (sh), 1435*m* (sh), 1380*m*, 1355*m*, 1325*m*, 1245*s*, 1220*w*, 1170*w*, 1100*m*, 1075*m*, 1015*s*, 1000*m*. 970*s*, 935*w*, 890*m*, 865*m*, 830*s*. <sup>1</sup>H-NMR (300 MHz, C<sub>6</sub>D<sub>6</sub>): 0.19, 0.22, 0.41, 0.49 (4*s*, 4 CH<sub>3</sub>Si); 0.99, 1.00 (2*s*, 2 CH<sub>3</sub>–C(6')); 1.05, 1.18 (2*s*, 6 CH<sub>3</sub>CSi); 1.24 1.28, 1.52–1.59 (2*m*, 2 H–C(5'), 2 H–C(5'')); 1.43 (*s*, 2 CH<sub>3</sub>–C(6'')); 1.73 (*m*,  $w_{1/2} \approx 3$ , CH<sub>3</sub>–C(2')); 1.96, 1.98 (2*d*,  $J \approx 1$ , CH<sub>3</sub>–C(2), CH<sub>3</sub>–C(6)); 2.02–2.15 (*m*, 2 H–C(4'), 2 H–C(4''), 2.07 (*m*,  $w_{1/2} = 6$ , CH<sub>3</sub>–C(2'')); 2.57 (*d*, J = 11, H–C(1')); 5.30 (*dm*, J = 11,  $w_{1/2} \approx 4$ , H–C(1)); 5.42 (*m*,  $w_{1/4} = 9$ , H–C(3'')); 5.71 (*m*,  $w_{1/2} = 9$ , H–C(3'')); 5.84 (*s*, H–C(3)); 6.15 (*s*, H–C(7)). <sup>13</sup>C-NMR (75 MHz, C<sub>6</sub>D<sub>6</sub>): –4.0, –3.8, –2.9, –2.6 (4*q*, 4 CH<sub>3</sub>Si); 11.3 (*q*, CH<sub>3</sub>–C(6'), CH<sub>3</sub>–C(2''), 2 CH<sub>3</sub>–C(6'')); 26.2, 27.6 (2*q*, 6 CH<sub>3</sub>CSi); 23.6 (2*t*, overlapping, C(5'), C(5''')); 33.3 (*t*, C(4')); 39.6 (*t*, C(4''')); 48.8 (*d*, C(1')); 10.1 (*d*, C(3)); 120.8 (*d*, C(3')); 125.6, 127.0, 129.3 (3*d*, C(1), C(7), C(3''')); 18.5, 19.4 (2*s*, 2 CSi); 33.2, 35.3 (2*s*, C(6'), C(6'')); 12.1, 129.2, 133.3, 135.4, 138.0, 142.7 (6*s*, C(2), C(5), C(6), C(2'', C(1''), C(2''')). MS: 612 (< 1, *M*<sup>+'</sup>, C<sub>38</sub>H<sub>68</sub>O<sub>2</sub>Si<sub>2</sub>), 555 (1), 481 (< ) 1), 309 (20), 308 (73), 307 (98), 305 (10), 251 (31), 249 (17), 186 (14), 185 (83), 175 (52), 159 (10), 147 (14), 127 (14), 119 (29), 115 (19), 105 (19), 75 (48), 74 (26), 73 (100), 69 (14), 59 (21). Anal. calc. for C<sub>38</sub>H<sub>68</sub>O<sub>2</sub>Si<sub>2</sub> (613.14): C 74.44, H 11.18; found: C 74.50, H 11.36.

2.3. Photolyses of (E)-9. 2.3.1. In THF. A soln. of (E)-9 (720 mg, 2.47 mmol) in abs. THF (150 ml) was irradiated ( $\lambda > 347$  nm, lamp B, filter A, ca. 100% conversion) under Ar for 3 h. CC (SiO<sub>2</sub>; hexane/Et<sub>2</sub>O 100:1) gave a mixture of dimers **31A** + **B** (199 mg, 28%) and intractable material.

2.3.2. In MeCN. A soln. of (E)-9 (1.045 g, 3.58 mmol) in abs. MeCN (170 ml) was irradiated as described above for 2.5 h (92% conversion). CC gave a mixture of dimers 31A + B (260 mg, 27%), and intractable material. Further purification of 31A + B by CC led to decomposition.

3. Thermolyses. – 3.1. Thermolyses of (E)- and (Z)-7. 3.1.1. Thermolysis of (E)-7. a) FVT (520°) of (E)-7 (187 mg, 0.702 mmol) in a silylated quartz tube without packing (100% conversion) afforded 32 (79 mg, 42%). b) FVT (350°) of (E)-7 (180 mg, 0.675 mmol) afforded 32 (44 mg, 32%) and (E/Z)-7 (ca. 5:1; 44 mg; 76% conversion).

(E)-*I*-[(tert-*Butyl*)*dimethylsilyloxy*]-7-*methyl*-3-*methyliden*-*I*,*6*-octadiene (**32**). B.p. 100°/0.07 Torr. UV (0.158 mg in 20 ml): 240 (20 300). IR: 3070w, 3030w, 2950s, 2920s, 2890s, 2850s, 1750w, 1635s, 1600m, 1465m (sh), 1460m, 1445m, 1415w, 1385w, 1370m, 1360m, 1330w, 1250s, 1185s, 1165s (br.), 1100w, 1000w, 920s, 895m, 870s. <sup>1</sup>H-NMR (300 MHz, C<sub>6</sub>D<sub>6</sub>): 0.07 (s, 2 CH<sub>3</sub>Si); 0.93 (s, 3 CH<sub>3</sub>CSi); 1.54, 1.65 (2*m*,  $w_{Y_2} = 4$ , CH<sub>3</sub>--C(7), 3 H--C(8)); 2.21 2.34 (*m*, 2 H--C(4), 2 H--C(5)); 4.81, 4.91 (2*d*,  $J_1 = 1.3, 0.9$ , CH<sub>2</sub>=-C(3)); 5.23 (*m*,  $w_{Y_2} = 9$ , H--C(6)); 6.45 (*AB*,  $J = 12, \delta_A = 6.13, \delta_B = 6.77$ , H--C(1), H--C(2)). <sup>13</sup>C-NMR (75 MHz, C<sub>6</sub>D<sub>6</sub>): -5.1 (*q*, 2 CH<sub>3</sub>Si); 17.8, 25.9 (2*q*, CH<sub>3</sub>--C(7), C(8)); 25.8 (*q*, 3 CH<sub>3</sub>CSi); 27.6, 33.5 (2*t*, C(5), C(4)); 111.0 (*t*, CH<sub>2</sub>=-C(3)); 116.1 (*d*, C(2)); 124.8 (*d*, C(6)); 141.8 (*d*, C(1)); 18.5 (*s*, CSi); 131.3 (*s*, C(7)); 144.0 (*s*, C(3)). MS: 266 (11,  $M^+$ , C<sub>16</sub>H<sub>30</sub>OSi), 224 (15), 223 (77), 209 (17), 186 (14), 150 (12), 141 (35), 135 (24), 129 (10), 107 (14), 75 (91), 73 (100), 69 (96), 59 (17), 43 (11), 41 (52).

3.1.2. Thermolysis of (Z)-7. FVT (520°) of (Z)-7 (87 mg, 0.326 mmol) as described above gave (E)-33 (40 mg, 46%).

 $(1E.3E)-1-[(tert-Butyl)dimethylsilyloxy]-3,7-dimethyl-1,3,6-octatriene ((E)-33). B.p.: 100°/0.07 Torr. UV (0.252 mg in 20 ml): 242 (18 800), 282 (2400). IR: 3030w, 2950s, 2920s, 2880s, 2850s, 1675w (br.), 1640s, 1615m, 1465m, 1460m, 1440m, 1405w, 1390m, 1375m, 1360m, 1295m, 1250s, 1215m, 1185m, 1170m, 1155s (sh), 1100m, 1055m, 1020w, 1005m, 980m, 935m, 920m, 870s. <sup>1</sup>H-NMR (300 MHz, C<sub>6</sub>D<sub>6</sub>): 0.09 (s, 2 CH<sub>3</sub>Si); 0.96 (s, 3 CH<sub>3</sub>CSi); 1.55 (s. CH<sub>3</sub>-C(7)); 1.65 (s, 3 H-C(8)); 1.71 (s, CH<sub>3</sub>-C(3)); 2.84 (t, J = 9, 2 H-C(5)); 5.20, 5.35 (2t, J = 9, H-C(4), H-C(6)); 6.35 (AB, J = 12.5, <math>\delta_A = 6.08, \delta_B = 6.62, H-C(1), H-C(2)$ . <sup>13</sup>C-NMR (75 MHz, C<sub>6</sub>D<sub>6</sub>): -5.04 (q, 2 CH<sub>3</sub>Si); 12.9, 17.8, 25.9 (overlapping with additional q) (3q, CH<sub>3</sub>-C(3), CH<sub>3</sub>-C(7), C(8), 3 CH<sub>3</sub>CSi); 27.5 (t, C(5)); 119.3 (d, C(2)); 123.6 (d, C(6)); 126.5 (d, C(4)); 140.0 (d, C(1)); 18.5 (s, CSi); 130.9, 131.3 (2s, C(3), C(7)). MS: 266 (1,  $M^{++}$ , C<sub>16</sub>H<sub>30</sub>OSi), 251 (1), 223 (1), 209 (7), 186 (10), 135 (7), 103 (18), 75 (100), 73 (26), 69 (19).

3.2. Thermolyses of (E)- and (Z)-8. 3.2.1. FVT (560°) of (E)-8 (733 mg, 2.4 mmol) in a silvlated quartz tube (81% conversion) followed by CC (hexane) afforded **34A** (124 mg, 17%) and a *ca.* 1:7 mixture (257 mg, 35%) of **34B**/35, which was separated by prep. GC (5% SE-30, 185°) affording **34B** (20 mg, 3%) and **35** (130 mg, 18%).

3.2.2. FVT of (Z)-8 (40 mg, 0.13 mmol) as described above gave 34A (6 mg, 15%), and a *ca.* 1:7 mixture (14 mg, 35%) of 34B/35.

3-[3'-(tert-Butyl)dimethylsilyloxy-2'-methyl-2'-propenylidene]-2,4,4-trimethylcyclohexene, Isomer A (**34**A). UV (0.186 mg in 10 ml): 221 (9390), 267 (9555). IR: 3010w, 2950s, 2920s, 2850s, 1645m. 1480m, 1435w (sh), 1385w, 1370w, 1355w, 1255m, 1185s, 1170s, 1155s, 1080w, 1000w, 935w, 895m, 865s, 835s. <sup>1</sup>H-NMR (300 MHz, C<sub>6</sub>D<sub>6</sub>): 0.04 (s, 2 CH<sub>3</sub>Si); 0.94 (s, 3 CH<sub>3</sub>CSi); 1.13 (s, 2 CH<sub>3</sub>C(4)); 1.48 (dd,  $J_1 = J_2 = 6.5$ , 2 H–C(5)); 1.89 (m,  $w_{J_2} = 3$ , CH<sub>3</sub>–C(2)); 2.04–2.14 (m, 2 H–C(6)); 2.07 (d, J = 1.5, CH<sub>3</sub>–C(2')); 5.48 (m,  $w_{J_2} = 8$ , H–C(1)); 5.86 (m,  $w_{J_2} = 5$ , H–C(1')); 6.32 (dq, J = 3, 1.5, H–C(3')). <sup>13</sup>C-NMR (75 MHz, C<sub>6</sub>D<sub>6</sub>): -4.9 (q, 2 CH<sub>3</sub>Si); 14.6 (q, CH–C(2')); 2.3.1 (q, CH<sub>3</sub>–C(2)); 25.9 (q, 3 CH<sub>3</sub>CSi); 27.5 (q, 2 CH<sub>3</sub>–C(4)); 24.2 (t, C(5)); 37.8 (t, C(6)); 121.9 (d, C(1)); 127.9 (d, C(1')); 138.0 (d, C(3')); 18.3 (s, CSi); 35.7 (s, C(4)); 116.9, 133.0, 145.5 (3s, C(2), C(3), (2')). MS: 306 (25,  $M^{++}$ , C<sub>19</sub>H<sub>34</sub>OSi), 291 (25), 235 (11), 175 (24), 173 (13), 159 (30), 119 (25), 75 (100), 73 (99), 59 (13), 41 (12). Anal. calc. for C<sub>19</sub>H<sub>34</sub>OSi (306.38): C 74.44, H 11.18; found: C 74.67, H 11.15.

*Isomer B* (34B), contaminated with *ca*. 20% of 35. IR: 3010w, 2950s, 2920s, 2850s, 1645m, 1460m, 1435w (sh), 1385w, 1370w. 1355w, 1255m, 1185s, 1170s, 1155s, 1180w, 1000w, 935w, 895m, 865s, 835s. <sup>1</sup>H-NMR (300 MHz, C<sub>6</sub>D<sub>6</sub>): 0.05 (s, 2 CH<sub>3</sub>Si); 0.95 (s, 3 CH<sub>3</sub>CSi); 1.17 (s, 2 CH<sub>3</sub>-C(4)); 1.54 (*dd*,  $J_1 = J_2 = 6.5$ , 2 H–C(5)); 1.69 (*m*,  $w_{V_2} = 3$ , CH<sub>3</sub>-C(2)); 2.05 (*m*,  $w_{V_2} = 4$ , CH<sub>3</sub>-C(2')); 2.08–2.13 (*m*,  $w_{V_2} = 14$ , C(6)); 5.45–5.47 (*m*,  $w_{V_2} = 8.5$ , H–C(1')); 5.82 (*m*,  $w_{V_2} = 5$ , H–C(1')); 6.01 (*dq*, J = 3, 1, H–C(3')). <sup>13</sup>C-NMR (75 MHz, C<sub>6</sub>D<sub>6</sub>): -5.0 (*q*, 2 CH<sub>3</sub>Si); 13.4 (*q*, CH<sub>3</sub>-C(2')); 22.15 (*q*, CH<sub>3</sub>-C(2)); 25.7 (*q*, 3 CH<sub>3</sub>CSi); 28.0 (2*q*, 2 CH<sub>3</sub>-C(4)); 24.3 (*t*, C(5)); 37.8 (*t*, C(6)); 119.1 (*d*, C(1)); 126.0 (*d*, C(1')); 135.0 (*d*, C(3')); 18.3 (*s*, CSi); 31.9 (*s*, C(4)); 116.8, 134.1, 146.8 (3*s*, C(2), C(3), C(2')). MS: 306 (23,  $M^{++}$ , C<sub>19</sub>H<sub>34</sub>OSi), 291 (20), 235 (14), 175 (25), 173 (16), 159 (35), 119 (26), 75 (100), 73 (91), 59 (12).

9-*[* (tert-*Butyl*)*dimethylsilyloxy*]-1,5,5,8-tetramethylbicyclo[4.3.0]nona-2,8-diene (**35**). IR: 3020*w*, 2950*s*, 2920*s*, 2850*s*, 1680*m*, 1465*m* (sh), 1457*m*, 1445*m* (sh), 1380*m*, 1360*m*, 1305*m*, 1280*w* (sh), 1265*s* (sh), 1255*s* (sh), 1245*s*, 1205*s*, 1150*w*, 1175*m*, 1115*m*, 930*w*, 880*s*, 835*s*. <sup>1</sup>H-NMR (300 MHz, C<sub>6</sub>D<sub>6</sub>): 0.18, 0.19 (2*s*, 2 CH<sub>3</sub>Si); 0.88, 1.34 (2*s*, 2 CH<sub>3</sub>-C(5)); 1.04 (*s*, 3 CH<sub>3</sub>CSi); 1.34 (*s*, CH<sub>3</sub>-C(1)); 1.52 (*dddd*,  $J_1 = 16$ ,  $J_2 = 5$ ,  $J_3 = J_4 = 1.5$ , H-C(4)); 1.56 (*dd*,  $J_1 = J_2 = 1$ , CH<sub>3</sub>-C(8)); 1.82 (br. *dd*,  $J_1 = J_2 = 8.5$ , H-C(6)); 1.90–2.12 (*m*, 2 H-C(7), H-C(4)); 5.56 (*ddd*, J = 10, 5.5, 2.5, H-C(3)); 5.90 (*dm*, J = 10,  $w_{V_2} = 6$ , H-C(2)). <sup>13</sup>C-NMR (75 MHz, C<sub>6</sub>D<sub>6</sub>): - 3.0 (*q*, 2 CH<sub>3</sub>Si); 13.4 (*q*, CH<sub>3</sub>-C(8)); 26.4 (3*q*, 3 CH<sub>3</sub>CSi); 26.0, 28.0, 29.6 (3*q*, CH<sub>3</sub>-C(1), 2 CH<sub>3</sub>-C(6)); 35.3, 35.7 (2*t*, C(4), C(7)); 51.9 (*d*, C(6)); 123.0, 132.6 (2*d*, C(2), C(3)); 19.0 (*s*, CSi); 31.9 (*s*, C(5)); 48.9 (*s*, C(1)); 107.7 (*s*, C(8)); 151.7 (*s*, C(9)). MS: 306 (22,  $M^{++}$ , C<sub>19</sub>H<sub>34</sub>OSi), 292 (25), 291 (100), 249 (12), 175 (12), 159 (10), 75 (37), 73 (92), 59 (16), 57 (11), 43 (10), 41 (18).

3.3. Thermolysis of (E)-9. FVT (600°) of (E)-9 (1.073 g, 3.5 mmol) as described in Sect. 3.2 afforded a mixture (883 mg) of starting material (E)-9 (23%), 36 (30%), and 37A + B (47%). This mixture was separated by prep. GC (5% SE-30, 185°) yielding 36 (55 mg) and 37A + B (148 mg).

9-[(tert-Butyl)dimethylsilyloxy]-1,5,5-trimethylbicyclo[4.3.0]nona-2,8-diene (**36**). IR: 3050w, 3010m, 2940s (sh), 2880s (sh), 1730w (br.), 1640s, 1455s, 1440m, 1380m, 1360s, 1315s, 1295s, 1280s, 1245s, 1230s, 1165m, 1105s, 990m, 960m, 935m (sh), 910s, 880s, 870s. <sup>1</sup>H-NMR (300 MHz, C<sub>6</sub>D<sub>6</sub>): 0.13, 0.15 (2s, 2 CH<sub>3</sub>Si); 0.88, 0.15 (2s, 2 CH<sub>3</sub>-C(5)); 0.98 (s, 3 CH<sub>3</sub>CSi); 1.35 (s, CH<sub>3</sub>-C(1)); 1.56 (ddd,  $J_1 = 17$ ,  $J_2 = 5.5$ ,  $J_3 = J_4 = 1.5$ , H-C(4)); 1.88 (dd br.,  $J_1 = J_2 = 8.5$ , H-C(6)); 1.97 (ddd, J = 17, 2.5, 1.5, H-C(4)); 2.09 (ddd, J = 15, 8, 2.1, H-C(7)); 2.15 (ddd,  $J_1 = 10$ ,  $J_2 = J_3 = 1.5$ , H-C(2)); 3.88 (dd,  $J_1 = 10$ ,  $Z_2 = J_3 = 1.5$ , H-C(2)); 3.88 (dd,  $J_1 = 10$ ,  $Z_2 = J_3 = 1.5$ , H-C(2)); 3.56 (22 MHz, C<sub>6</sub>D<sub>6</sub>): -4.5 (q, 2 CH<sub>3</sub>Si); 26.0 (q, 3 CH<sub>3</sub>CSi) and presumably 25.6, 28.2 (3q, 2 CH<sub>3</sub>-C(5), CH<sub>3</sub>-C(1)); 29.5, 35.6 (2t, C(4), C(7)); 53.0 (d, C(6)); 97.2 (d, C(8)); 123.5, 132.4 (2d, C(2), C(3)); 18.4 (s, CSi); 31.8 (s, C(5)); 48.0 (s, C(1)); 159.8 (s, C(9)).

 $\begin{array}{l} (E) -3 - \left[ 2' - (3'', 3'', 4'', 4'' - Tetramethyl - 2'' - oxa - 3'' - silacyclopentyl) vinyl \right] - 2, 4, 4-trimethylcyclohexene \quad (37A + B), \\ 1 : 1 \ Mixture. 1R: 3020w, 2940s, 2900s, 2850s, 1455m, 1430m (sh), 1380m, 1370m (sh), 1360m, 1245s, 1160w, 1070m (br.), 1020s, 1000s, 965m, 935m, 890s, 850s. ^{1}H - NMR (300 MHz, C_6D_6): 0.08, 0.09, 0.11 (3s, 2 CH_3Si); 0.90, 0.93 (3 H); 0.97, 0.98 (4s, 2 CH_3 - C(4''), 2 CH_3 - C(4)); 1.10 - 1.16 (m, H - C(5)); 1.46 - 1.54 (m, 2 H - C(5''), H - C(5)); 1.70, \\ 1.72 (2m, w_{V_2} = 5, CH_3 - C(2)); 1.97 (m, w_{V_2} = 15, 2 H - C(6)); 2.13 (d, J = 7, H - C(3)); 4.44 - 4.46 (m, H - C(1'')); \\ 5.41 (m, w_{V_2} = 8, H - C(1)); 5.50 - 5.70 (m, H - C(1'), H - C(2')). ^{13}C - NMR (25.2 MHz, C_6D_6): -4.5, -2.6 (2q, 2 CH_3 Si); \\ 22.1 (q, CH_3 - C(2)); 22.9, 23.2 (2q, 2 CH_3 - C(4'')); 26.1, 26.6 (2q, 2 CH_3 - C(4)); 22.4 (t, C(5)); 30.9 (t, C(6)); 49.5 (t, C(5'')); 53.2 (d, C(3)); 74.9 (d, C(1'')); 120.0 (d, C(1)); 129.3, 134.7 (2d, C(2'), C(1')); 21.5 (s, C(4'')); \\ 31.1 (s, C(4)); 133.4 (s, C(2)). MS: 292 (4, M^{++}, C_{18}H_{32}OSi), 236 (21), 221 (19), 180 (11), 170 (16), 169 (100), 165 (16), 123 (15), 113 (86), 105 (15), 91 (17), 77 (10), 75 (86), 59 (12), 41 (13). \\ \end{array}$ 

**4.** Additional Experiments. - 4.1. *Hydrolysis of* 27A + B. A soln. of 27A + B (104 mg, 0.195 mmol) in Et<sub>2</sub>O (5 ml) was stirred with 2N aq. HCl (1 ml) overnight. The mixture was worked up in Et<sub>2</sub>O, and CC (hexane/Et<sub>2</sub>O 30:1) afforded (E/Z)-7 (14 mg, 27%), (E/Z)-10 (28 mg, 94%), and 28 (36 mg, 69%).

4.2. Hydrolysis of **30A** and **30B**. a) A soln. of **30A** (77 mg, 0.125 mmol) in Et<sub>2</sub>O (10 ml) was stirred with 5% aq. HCl (5 ml) for 72 h at r.t. The mixture was worked up with Et<sub>2</sub>O, and CC (Et<sub>2</sub>O/hexane 1:50) afforded (Z)-8 (15

4.3. Hydrolysis of 31A + B. A soln. of 31A + B (260 mg, 0.45 mmol) in Et<sub>2</sub>O (15 ml) and 5% aq. HCl (5 ml) was stirred for 24 h at r.t. The mixture was worked up in Et<sub>2</sub>O, and CC (Et<sub>2</sub>O/pentane 1:4) yielded 12 (83 mg, 52%).

4.4. Hydrolysis of (E)-8 and (E)-9. a) A soln. of (E)-8 (19 mg, 0.06 mmol) in Et<sub>2</sub>O (3 ml) was stirred with 10% aq. HCl (1 ml) for 24 h at r.t. Workup in Et<sub>2</sub>O and CC (Et<sub>2</sub>O/hexane 1:3) gave 11 (10 mg, 83%). b) Hydrolysis of (E)-9 (18 mg, 0.06 mmol) in Et<sub>2</sub>O (3 ml) with 5% aq. HCl (1 ml) for 24 h at r.t. afforded, after CC, 12 (10 mg, 88%).

4.5. Synthesis of **32** and (E/Z)-**33**. Citral ((E/Z)-**10**; 1.00 g, 6.57 mmol; freshly distilled) was added to a soln. of (*t*-Bu)Me<sub>2</sub>SiCl (1.18 g, 7.86 mmol) in abs. Et<sub>3</sub>N (2.2 ml, 1.59 g, 15.76 mmol) and abs. DMF (1.7 ml), which had been filtered through *Celite*. The mixture was heated at reflux temp. for 3 d, then cooled to 0°, diluted with pentane, and washed with 0.1N aq. HCl. CC (hexane/Et<sub>2</sub>O 10:1) gave a mixture of **32** and (E/Z)-**33** (668 mg, 92%; at 42% conversion; ratio *ca*. 1:1:1). Repeated CC (hexane) afforded an anal. sample of **32** and a 5:1 mixture of (E/Z)-**33**.

 $(1E_3Z)$ -f( tert-Butyl) dimethylsilyloxy f-3,7-dimethyl-1,3,6-octatriene ((Z)-33; ca. 1:5 mixture with (E)-33). B.p.: 100°/0.07 Torr. Characteristic signals assigned to (Z)-33: <sup>1</sup>H-NMR (300 MHz, C<sub>6</sub>D<sub>6</sub>): 0.09 (s, 2 CH<sub>3</sub>Si); 0.94 (s, 3 CH<sub>3</sub>CSi); 1.53 (s, CH<sub>3</sub>-C(7)); 1.62 (m,  $w_{y_2} = 4$ , 3 H-C(8)); 1.81 (m,  $w_{y_2} = 4$ , CH<sub>3</sub>-C(3)); 2.88 (t, J = 7, 2 H-C(5)); 5.23 (m,  $w_{y_2} = 9$ , H-C(4), H-C(6)); 6.58 (AB, J = 12,  $\delta_A = 6.47$ ,  $\delta_B = 6.69$ , H-C(2), H-C(1)). <sup>13</sup>C-NMR (75 MHz, C<sub>6</sub>D<sub>6</sub>): -5.0 (q. 2 CH<sub>3</sub>Si); 20.9, 25.9 (overlapping with additional q), 30.3 (3q, 3 CH<sub>3</sub>CSi, CH<sub>3</sub>-C(3), CH<sub>3</sub>-C(7), C(8)); 27.1 (t, C(5)); 112.0 (d, C(2)); 123.8 (d, C(6)); 125.6 (d, C(4)); 142.4 (d, C(1)); 18.5 (s, CSi); 129.6, 131.5 (2s, C(3), C(7)).

4.6. *Hydrolysis of* **35**. A soln. of **35** (20 mg, 0.065 mmol) in THF (5 ml), and 5% aq. HCl (2 ml) was stirred for 3 h at r.t. The mixture was worked up with  $Et_2O$ , and CC ( $Et_2O$ /hexane 1:10) gave **38** (12 mg, 95%).

2.2,6,8-Tetramethylbicyclo[4.3.0]non-4-en-7-one (**38**). IR: 3020m, 2950s, 2920s, 2860s, 2830m, 1645w, 1730s, 1470m, 1445m, 1380m, 1370w, 1360m, 1345w, 1325w, 1280w, 1260w, 1240w, 1195m, 1165w, 1145w, 1105w, 1040w, 990m, 955w, 925w, 895w, 705s. <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>): 1.00, 1.02 (2s, 2 CH<sub>3</sub>-C(2)); 1.08 (*d*, J = 6, CH<sub>3</sub>-C(8)); 1.25 (s, CH<sub>3</sub>-C(6)); 1.68 (*dm*, J = 18, 5.5,  $w_{1/2} = 3$ , H-C(3)); 1.74 (br. *dd*, J = 11.5, 6, H-C(1)); 1.91 (*dm*, J = 18,  $w_{1/2} = 6$ , H-C(3)); 2.10-2.30 (*m*, H-C(8), H-C(9)); 5.32 (*ddm*, J = 10, 2.5,  $w_{1/2} = 3$ , H-C(5)); 5.61 (*ddd*, J = 10, 5.5, 2, H-C(4)). MS: 192 (16,  $M^{++}$ , C<sub>13</sub>H<sub>20</sub>O), 135 (14), 134 (64), 122 (22), 121 (19), 108 (12), 107 (100), 93 (11), 91 (24), 79 (10), 77 (10), 41 (15).

4.7. Hydrolysis of **36**. A soln. of **36** (36 mg, 0.12 mmol) in THF (6 ml) was stirred with 5% aq. HCl (2.5 ml) for 1 h. Workup in Et<sub>2</sub>O and CC (Et<sub>2</sub>O/hexane 1:1) yielded **39** (20 mg, 88%).

2.2,6-Trimethylbicyclo[4.3.0]non-4-en-7-one (**39**). IR: 3010*m*, 2950*s*, 2920*s*, 2860*s*, 2830*m*, 1730*s*, 1640*w*, 1470*m*, 1445*s*, 1430*m*, 1405*s*, 1380*s*, 1360*s*, 1345*m*, 1320*w*, 1295*m*, 1270*m*, 1250*m*, 1225*m*, 1205*m* (sh), 1190*s*, 1140*w* (sh), 1130*m*, 1075*s*, 1050*m*, 1030*m*, 1015*m*, 990*m*, 965*m*, 945*m*, 900*m*, 860*w*, 830*w*. <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>): 0.97, 1.03 (2*s*, 2 CH<sub>3</sub>-C(2)); 1.23 (*s*, CH<sub>3</sub>-C(6)); 1.45 -1.53 (*m*, H-C(9)); 1.74 (*ddm*, *J* = 18, 5.5, *w*<sub>1/2</sub> = 3, H-C(3)); 1.82 (br. *dd*, *J* = 10.5, 7, H-C(1)); 1.94 (*dm*, *J* = 18, *w*<sub>1/4</sub> = 6, H-C(3)); 1.92-2.06 (*m*, H-C(9)); 2.18 (*ddd*, *J* = 18.5, 11.5, 9, H-C(8)); 2.34 (*ddd*, *J* = 18.5, 8.5, 2, H-C(8)); 5.33 (*ddm*, *J* = 10, 2.5, *w*<sub>1/4</sub> = 3, H-C(5)); 5.65 (*ddd*, *J* = 10, 5.5, 2.5, H-C(4)). MS: 178 (24,  $M^{++}$ , C<sub>12</sub>H<sub>18</sub>O), 135 (13), 134 (77), 122 (19), 121 (35), 120 (16), 119 (60), *107* (100), 105 (14), 94 (13), 93 (14), 91 (32), 79 (23), 77 (16), 65 (11), 55 (10), 41 (28).

4.8. *Hydrolysis of* **37A** + **B**. A soln. of **37A** + **B** (36 mg, 0.12 mmol) in THF (6 ml) and 5% aq. HCl (3 ml) was heated under reflux for 3 h. Workup in Et<sub>2</sub>O and CC (Et<sub>2</sub>O/hexane 1:10) gave **40** (11 mg, 32%).

 $(4 \text{ E}, 6 \text{ E})^{-7-(2', 6', 6'-Trimethyl-2'-cyclohexenyl)^{-2,3,3-trimethyl-2-sila-4,6-heptadien-2-ol}$  (40). UV (0.156 mg in 10 ml): 250 (28 090). IR : 3680m, 3010m, 2950s, 2920s, 2860s, 1460m, 1380m, 1360m, 1300m, 1250s, 1145w (br.), 1075w, 990s, 965m, 940w, 890m (sh). <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>): 0.10 (s, 2 CH<sub>3</sub>Si); 0.82, 0.89 (2s, 2 CH<sub>3</sub>-C(6')); 1.06 (s, 2 CH<sub>3</sub>-C(3)); 1.18-1.45 (m, 2 H-C(5')); 1.56 (br. s, OH); 1.59 (m,  $w_{V_2} = 5$ , CH<sub>3</sub>-C(2')); 1.99 (m,  $w_{V_2} = 15$ , 2 H-C(4')); 2.08 (d, J = 9.5, H-C(1')); 5.36 (dd, J = 15, 9.5, H-C(7)); 5.38 (m,  $w_{V_2} = 6$ , H-C(3')); 5.65 (d, J = 15, H-C(4')); 5.88 (dd, J = 14.5, 10, H-C(5)); 5.96 (dd, J = 15, 10, H-C(6)). <sup>13</sup>C-NMR (75 MHz, C<sub>6</sub>D<sub>6</sub>): -3.4 (2q, 2 CH<sub>3</sub>Si); 22.2 (2q, 2 CH<sub>3</sub>-C(3)); 23.2 (q, CH<sub>3</sub>-C(2')); 27.0, 27.8 (2q, 2 CH<sub>3</sub>-C(6')); 23.4 (r, C(5')); 31.8 (r, C(4')); 54.8 (d, C(1')); 120.9 (d, C(3')); 126.6, 132.0, 132.9, 140.3 (4d, C(4), C(5), C(6), C(7)); 28.0 (s, C(3)); 32.5 (s, C(6')); 134.5 (s, C(2')). MS: 292 (8,  $M^{++}$ , C<sub>18</sub>H<sub>32</sub>OSi), 169 (53), 123 (25), 95 (20), 84 (28), 78 (16), 77 (17), 75 (100), 69 (20), 56 (41), 55 (23), 44 (22), 42 (22), 41 (42).

## REFERENCES

- [1] A. O'Sullivan, B. Frei, O. Jeger, Helv. Chim. Acta 1986, 69, 555.
- [2] M.E. Scheller, Diss. ETH No. 7896, 1985.
- [3] M.E. Scheller, B. Frei, Helv. Chim. Acta 1984, 67, 1734.
- [4] H.D. Hartzler, 'Unsaturated Carbenes', in 'Carbenes', Eds. R.A. Moss and M. Jones, John Wiley, New York, 1975, Vol. 2, pp. 57.
- [5] R. L. Danheiser, D. M. Finck, K. Okano, Y.-M. Tsai, S. W. Szepanski, J. Org. Chem. 1985, 50, 5393.
- [6] a) J.P. O'Brien, A.I. Rachlin, S. Teitel, J. Med. Chem. 1969, 9, 1112; b) A. Hoppmann, P. Weyerstrahl, W. Zummack, Liebigs Ann. Chem. 1977, 1547.
- [7] E. Fujita, Y. Nagao, K. Kaneko, Chem. Pharm. Bull. Jpn. 1976, 24, 1115.
- [8] T. Oritani, K. Yamashita, Agric. Biol. Chem. 1970, 34, 830.
- [9] V. Rautenstrauch, Helv. Chim. Acta 1973, 56, 2492.
- [10] a) D. Seebach, Synthesis 1969, 17; b) E.J. Corey, D. Seebach, R. Freedman, J. Am. Chem. Soc. 1967, 89, 434.
- [11] A. O'Sullivan, N. Bischofberger, B. Frei, O. Jeger, Helv. Chim. Acta 1985, 68, 1089.
- [12] M. Yoshioka, K. Ishii, H. R. Wolf, Helv. Chim. Acta 1980, 63, 571.
- [13] a) J. H. Noggle, R. E. Schirmer, 'The Nuclear Overhauser Effect', Academic Press, New York, 1971; b) J. D. Mcrsh, J. K. M. Sanders, Org. Magn. Reson. 1982, 18, 122.
- [14] R. Ricard, P. Sauvage, C. S. K. Wan, A. C. Weedon, D. F. Wong, J. Org. Chem. 1986, 51, 62.
- [15] A.G. Brook, Acc. Chem. Res. 1974, 7, 77, and ref. cited therein.
- [16] J. K. Crandall, C. F. Mayer, J. Org. Chem. 1970, 35, 3049.
- [17] R.A. Bourque, P.D. Davis, J.C. Dalton, J. Am. Chem. Soc. 1981, 103, 697.
- [18] J. Becker, J. Ehrenfreund, O. Jeger, G. Ohloff, H. R. Wolf, Helv. Chim. Acta 1974, 57, 2679.
- [19] C. Shih, J.S. Swenton, J. Org. Chem. 1982, 47, 2668.
- [20] G. Ohloff, Tetrahedron Lett. 1960, 10.
- [21] a) J. M. Conia, P. Le Perchec, Synthesis 1975, 1; b) K. Ishii, H. R. Wolf, O. Jeger, Helv. Chim. Acta 1980, 63, 1520.
- [22] A. R. Bassindale, A. G. Brook, J. Harris, J. Organomet. Chem. 1975, 90, C6-C8.